Total Synthesis of $(-)$ -Salicylihalamide

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Abstract: A concise total synthesis of the potent cytotoxic marine natural products salicylihalamide A and B $(1a, b)$ is reported. Key steps of our approach were the asymmetric hydrogenation reactions of β -keto esters 18 and 32 catalyzed by $[(S)$ -BINAP)Ru- Cl_2 , NEt₃ and the cyclization of the macrolide core by ring closing olefin metathesis (RCM) using the "secondgeneration∫ ruthenium carbene complex 24 as the catalyst which bears an

imidazol-2-ylidene ligand. The E/Z ratio obtained in this macrocyclization reaction was determined by the protecting groups at the remote phenolic OH group of the cyclization precursor. The elaboration of the resulting cycloalkene 37 into the final target involved a $CrCl₂$ -

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mediated synthesis of vinyliodide 49 which, after deprotection, did undergo a copper-catalyzed cross-coupling process with the (Z, Z) -configurated carboxamide 42 to form the labile enamide moiety of 1. Compound 42 was derived from a palladium-catalyzed Negishi coupling between butynylzinc chloride and 3-iodoacrylate 39 followed by a Lindlar reduction of enyne 40 thus obtained and a final aminolysis of the ester group.

Introduction

Bioassay-guided fractionation of the extracts of an unidentified sponge of the Haliclona genus collected off the Southwestern Australian coast led to the discovery of two novel macrolide antibiotics called salicylihalamide $A(1a)$ and B $(1b)$.^[1, 2] These compounds exhibited remarkably potent cytotoxicity in the 60-cell-line human tumor assay from the National Cancer Institute (NCI), with a mean $GI₅₀$ concentration of only about 15 nm. The melanoma cell lines showed the highest average sensitivity $(GI_{50} = 7 \text{ nm}, TGI = 60 \text{ nm}).$ Most notably, however, the activity profile of salicylihalamide in this assay showed no significant correlation to other compounds in the NCI database, indicating a novel mechanism of action.[1] From subsequent biochemical studies it was concluded that 1 does inhibit mammalian vacuolar-type (H⁺)-ATPases (V-ATPases) with an unprecedented selectivity; this suggests that these proton-translocating pumps may constitute a novel molecular target for cancer therapeutic agents.[3]

The most striking structural motif of 1 is the labile enamide linkage connecting a polyunsaturated domain with a salicylic acid derived macrolactone core. Since the discovery of 1a, b in 1997, several closely related natural products have been reported, including oximidine I (2) ,^[4] apicularene A (3) ,^[5]

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Salicylihalamide A (1a): 17(E) Salicylihalamide B (1b): 17(Z)

lobatamide A (4) , ^[6] or CJ-12950 (5) .^[7] In addition to the obvious constitutional and topological resemblance, these novel enamides also show pronounced cytotoxic activities against various human cancer cell lines and share with 1 a common mode of action.[3]

While the promising biological properties render salicylihalamide and congeners relevant targets for total synthesis, their unusual structural features provide an excellent forum for the validation of new synthetic methods. In this context, a truncated version of 1 was synthesized using olefin metathesis for the cyclization of the 12-membered ring and its biological activity has been evaluated.[8] Our study has provided strong evidence that the cytotoxicity of 1 is intimately related to the presence of an intact enamide bond and has also clearly featured the relevance of ring closing metathesis (RCM) as a strategic manoeuvre en route to this family of natural products. In fact, RCM later became the key transformation in all preparative approaches towards 1 and analogues thereof reported so far, including the first total syntheses of this rewarding target.^[9-11] A particularly noteworthy spin-off of

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these studies is the fact that the absolute stereochemistry of salicylihalamide had to be revised and is now assigned as it appears in structure $1 (12S, 13R, 15S)$.^[9a, 12]

Described below is a full account of our work in this field. Specifically, we outline a concise and inherently flexible approach to the core segment of 1 which relies on asymmetric hydrogenation reactions for the formation of the chiral centers. The efficient cyclization of the macrolide ring by RCM lends further credence to the notion that ™secondgeneration∫ ruthenium carbene complexes are superior catalysts for this type of transformation. Finally, the enamide moiety of the target was formed by a copper-catalyzed cross coupling process which delivered this rather labile structural motif in good yield.

Results and Discussion

Retrosynthetic analysis: Anticipating the formation of the macrolide ring by RCM,^[13] salicylihalamide can be deconvoluted into three synthons as shown in Scheme 1. Among the different possible disconnections for the enamide linkage, [14] we opted for a cross-coupling approach aiming at the direct formation the C $-N$ bond, which led back to a suitably functionalized vinyl halide core and an unsaturated amide side chain as the precursors. This strategy expands on model studies that were recently disclosed;[15] it holds the promise to be straightforward and flexible and may therefore outperform the methodology previously used en route to 1. [9] Since the enamide certainly constitutes a fragile linkage, its formation was postponed to the very end of the synthesis. Preferentially, it should be installed in the presence of the free OH groups because any deprotection step after the enamide formation might destroy this labile moiety. It was by no means clear at the beginning of this study whether the envisaged crosscoupling methodology would meet this serious constraint.

Although the aliphatic segment carrying the chiral centers of the molecule is certainly accessible by various routes, the application of asymmetric hydrogenation reactions seemed to be particularly appealing.^[16] This strategy can i) deliver the target by two iterative cycles of β -keto ester formations/

Scheme 1. Retrosynthetic analysis of salicylihalamide 1.

reductions, ii) promises reagent control over the absolute stereochemistry of the chiral centers formed, and hence iii) provides access to all possible diastereoisomers of 1, if desirable, simply by changing the chiral ligands to the hydrogenation catalysts.

Preparation of the salicylic acid part: Two different syntheses of the required salicylic acid part of $1a$, b have been developed. The first one used cheap 2,6-dihydroxybenzoic acid (6) as the starting material which was converted on a multigram scale into triflate 8 by formation of the isopropylidene derivative $7^{[17]}$ and subsequent reaction with triflic anhydride under standard conditions (Scheme 2).[18] This compound was allylated in high yield by a modified Suzukitype reaction^[19] according to a procedure previously developed in this laboratory.[20] Specifically, 9-allyl-9-BBN was treated with KOMe to afford a mixture of borate complexes

Scheme 2. [a] Acetone, SOCl₂, DMAP, DME, 96%; [b] triflic anhydride, pyridine, 85% ; [c] 9-allyl-9-BBN, KOMe, cat. [PdCl₂(dppf)], THF, 83% ; [d] BCl₃, CH₂Cl₂, 96%. BBN = 9-borabicyclo[3.3.1] nonane, dppf = Fe(η^5 - $C_5H_4PPh_2)_2$, DMAP = dimethylaminopyridine, DME = dimethoxyethane.

which rapidly transfer the allyl group to the triflate in the presence of catalytic amounts of $[PdCl_2(dppf)]$. Subsequent cleavage of the isopropylidene moiety of 9 was best achieved with BCl₃ in CH₂Cl₂ at 0° C, affording the desired salicylic acid 10 in almost quantitative vield.

Alternatively, the corresponding methyl ether 12 can be conveniently prepared from commercial 2-methoxybenzoic acid (11) by ortho-metalation using sBuLi followed by trapping of the resulting aryllithium species with allyl bromide at low temperature (Scheme 3).[21] Although the yield of this reaction amounted only to 48%, this approach was appealing in terms of its unrivaled "economy of steps".^[22]

Scheme 3. Synthesis of the required salicylic acid part by ortho-metalation.[21]

Model studies on RCM: At the outset of our project, the enantiomer of 1 was targeted because of the incorrect assignment of the absolute stereochemistry of salicylihalamide in the original publication (see above).^[1, 12] Therefore, the model studies reported below concerning the crucial RCM step have been conducted in the wrong enantiomeric series.

Our synthesis began with an asymmetric alkylation reaction of the oxazolidinone derivative 13 with prenyl bromide (Scheme 4), delivering product 14 in good yield.[23] Hydrolytic cleavage of the chiral auxiliary in the presence of $H_2O_2^{[24]}$ afforded acid 15 which was converted into acid chloride 16 under strictly neutral conditions using the chloroenamine reagent 17.^[25] Reaction of crude 16 with the lithium enolate of methyl acetate at low temperature provided β -keto ester $\mathbf{18}$, $^{[26]}$ which did undergo a ligand controlled asymmetric hydrogenation using $[(R)$ -BINAP)RuCl₂]₂ • NEt₃ as the catalyst (4 atm H₂, 80 °C).^[16, 27] This carbonyl reduction occured with almost perfect diastereoselectivity (de >99%). In line with literature precedence,[27a] the trisubstituted double bond of the substrate remained fully intact under these conditions.[28]

We were well aware, however, that metathesis catalysts such as the classical Grubbs ruthenium carbene complex $23^{[29]}$ are rather sensitive to the substitution pattern of the olefin.[30] Therefore the trisubstituted alkene in 22 which was necessary to impose a chemoselective path on the hydrogenation,[28] would likely impede the projected macrocyclization by RCM.

To evaluate this aspect prior to launching the actual total synthesis program, a model study was necessary. For this purpose, the secondary hydroxyl group in compound 19 was silylated and the resulting product 20 was reduced with $LiBEt₃H$ to afford primary alcohol 22 a. Subsequent esterification with acid 10 under Mitsunobu conditions^[31] afforded diene 25 a which allowed to test the crucial RCM step.

In fact, the reluctance of complex 23 to react with highly substituted alkenes^[30] was responsible for the failure in converting diene 25 into the desired macrocyclic product 26. It has recently been shown, however, that the exchange of one

Scheme 4. [a] i) LiHMDS, THF, -78° C, 30 min; ii) dimethylallyl bromide, 0° C, 16 h, 85%; [b] LiOH, H₂O₂, THF/H₂O, 0° C, 99%; [c] 17, CH₂Cl₂, 90 min; [d] i) LDA, methyl acetate, THF, -78 °C, 1 h; ii) addition of crude 16, RT, 2 h, 81%; [e] $[(R)$ -BINAP)RuCl₂]₂ · NEt₃ (0.8 mol%), H₂ (4 atm), MeOH, 80° C, 4 h, 96% ; [f] TBSOTf, 2,6-lutidine, CH₂Cl₂, 0[°]C, 1 h, 95%; [g] MOMCl, iPr_2NEt , DMAP cat., CH_2Cl_2 , 40 h, 90%; [h] $LiBEt_3H$, THF, RT, 83% (R = TBS), 98% (R = MOM); [i] DEAD, PPh₃, Et₂O, 96% (25a), 81% (25b). DEAD = diethyl azodicarboxylate, $MOMCl =$ methoxymethyl chloride.

 PCy_3 ligand in 23 for a N-heterocyclic carbene up-graded the performance of the catalyst to a significant extent and rendered the catalyst suitable for the cyclization of tri- and even tetrasubstituted cycloalkenes.[32] Gratifyingly, reaction of diene 25 with the "second-generation" ruthenium carbene complex 24 in toluene at 80° C afforded the 12-membered ring 26 in excellent yield (Scheme 5).^[33] While this result suggested that the envisaged cyclization of the macrolide core of salicylihalamide itself was feasible, we were surprised to find that the product was formed as a single isomer which was assigned the (Z)-configuration based on a careful analysis of its NMR data.

This stereochemical outcome was certainly unexpected for the following reasons:

i) The vast majority of RCM-based macrocylizations reported in the literature provided (E, Z) -mixtures, with the (E) isomer usually prevailing;[13]

Scheme 5. Model study concerning the formation of the core segment by RCM.

- ii) This trend pertained to our previous approach to a truncated salicylihalamide core which was obtained in a ratio of $E:Z = 2.3:1;^{[8]}$
- iii) NHC-containing metathesis catalysts were recently shown to be particularly (E) -selective;^[34, 35]
- iv) Complex 24, when applied to the total synthesis of zearalenone 29, a fungal metabolite that is closely related to the salicylihalamide core in structural terms, led to the exclusive formation of the desired (E) -isomer in excellent yield.[35a]

Since it is mandatory to form the (E) - rather than the (Z) isomer en route to 1, we studied whether the stereochemical outcome of the RCM reaction can be rectified (see Table 1). A simple means might be the use of "participating" protecting groups.[36] Unfortunately, however, replacement of the TBS group by a more strongly ligating methoxymethyl (MOM) ether had no effect on the course of the reaction, delivering (Z) -26b as the only product of RCM in 72% yield. Similarly, inversion of the stereocenter at C-13 of the cyclization precursor (easily achieved by reducing ketoester 18 with a ruthenium catalyst containing (S) -BINAP rather than (R) -BINAP as the ligand) did not change the outcome of the reaction; thus, cyclization of diene 27 again led to the exclusive formation of the (Z) -configurated product 28.^[37]

In trying to rationalize this surprising yet seemingly invariable stereochemical course, we speculated whether a hydrogen bond between the phenolic OH and the COOR

Table 1. Model study on RCM. All reactions were carried out using complex 24 (5 mol%) as the catalyst in toluene at 80° C unless stated otherwise.

Entry	Substrate	\mathbf{R}^1	\mathbb{R}^2	Product	Yield $[\%]$	E:Z
-1	25 a	TBS	H	26 a	89	0:100
2	25 _b	MOM	H	26 _b	$72^{[a]}$	0:100
3	25 c	MOM	TBS	26 c	97	30:70
$\overline{4}$	27	TBS	н	28	$80^{[a]}$	0:100

[a] Using 7.5 mol% of the catalyst.

group in the substrates prevented free rotation in this part of the molecule and hence enforced a strongly preferred conformation of the cyclization precursors. While addressing this issue, it was noticed that protection of the phenolic OH as a TBS ether further increased the efficiency of the cyclization (97% yield) and led to appreciable amounts of the desired (*E*)-isomer **26 c**. Although the $E:Z = 30:70$ ratio was far from optimal, this result suggested that it might be possible to approach salicylihalamide 1 by RCM, if the substituent at the remote phenolic OH group was properly chosen.[38]

Synthesis of the fully functional core of salicylihalamide: Based on the insights gained in this model study and learning from the literature that the absolute stereochemistry of 1 had originally been assigned incorrectly, $[9, 12]$ we embarked into the total synthesis of this promising antitumor agent.

The first steps simply translate the sequence outlined above into the other enantiomeric series. For this purpose, however, it turned out advantageous to use Oppolzer's bornanesultam auxiliary to control the diastereoselectivity of the prenylation step.^[39] Since compound 31 is crystalline (while the corresponding oxazolidinone derivative 14 is not) the purification of the crude product and hence the up-scaling of the reaction to multigram amounts was greatly facilitated. Saponification of 31, conversion of the resulting acid ent-15 into the acid chloride ent-16 as described, followed by chain extension with lithio methyl acetate provided β -oxo ester *ent*-18 which did undergo a chemo- and stereoselective hydrogenation in the presence of $[(S)$ -BINAP)RuCl₂ $]_2$ • NEt₃ as the catalyst (4 atm $H₂$, 80 °C). Protection of the resulting OH function in *ent*-19 by a MOM group sets the stage for the iterative construction of the remaining stereocenter. For this purpose, a chain extension was carried out using the lithium enolate of tertbutyl acetate as the reagent, followed by a diastereoselective reduction of the resulting β -keto ester 32 using the same catalyst, that is $[(S)$ -BINAP)RuCl₂l₂ · NEt₃. This hydrogenation provided best results under slightly modified conditions by lowering the temperature to 25° C but increasing the hydrogen pressure to 80 atm. Product 33 thus formed (de $>98\%$) was reduced with LiAlH₄ and the resulting diol 34 was converted into the mono-PMB ether derivative 35 by double deprotonation with excess NaH followed by addition of one equivalent of PMBCl; under these conditions, the alkylation of the primary alkoxide was strongly favored over the competing protection of the secondary one.^[40] Esterification of alcohol 35 with acid 10 under Mitsunobu conditions gave the desired ester 36 which served as a substrate for the envisaged macrocyclization.

In line with the observations made in the model studies, the stereochemical outcome of RCM depends on the remote substituents on the phenolic OH group (Table 2). Thus, the unprotected derivative 36 a led to the exclusive formation of the (Z) -configurated product 37a, the TBS-protected analogue **37b** afforded a $E:Z = 40:60$ mixture of both isomers, whereas the (E) -isomer prevailed when a methyl- $(37c)$ or a MOM group (37 d) blocked the phenolic site. We are unaware of any precedence, in which the choice of a remote protecting group allowed to alter the course of RCM to a similar extent.[38]

Scheme 6. [a] Lithium (cyclohexyl)(isopropyl)amide, THF, -78° C, 60 min; then dimethylallyl bromide, HMPA, $-78^{\circ}\text{C} \rightarrow \text{RT}$, 79%; [b] LiOH, H_2O_2 , THF/H₂O 4:1, 24 h, 95%; [c] chloroenamine 17, CH₂Cl₂, 90 min; [d] lithio methyl acetate, THF, $-78\degree C \rightarrow RT$, 81%; [e] [((S)-BINAP)RuCl₂]₂ • NEt₃ (0.4 mol%), H₂ (4 atm), MeOH, 80 °C, 4 h, 94%; [f] MOMCl, iPr_2NEt , cat. DMAP, CH_2Cl_2 , 40 h, 89%; [g] lithio tert-butyl acetate, THF, -40° C, 3 h, 95%; [h][((S)-BINAP)RuCl₂]₂ · NEt₃ (1.2 mol\%) , H₂ (80 atm), MeOH, 25 °C, 6.5 h, 93%; [i] LiAlH₄, Et₂O, 0°C, 6 h, 90%; [j] NaH, PMBCl, DMF, 90 min, 76%; [k] acid 12, PPh₃, DEAD, Et_2O , 20 h, 93%. PMB-Cl = *para*-methoxybenzyl chloride.

Table 2. RCM-based cyclization of diene 36 to cycloalkene 37. All reactions were carried out using complex 24 (5 mol%) as the catalyst in toluene at 80°C unless stated otherwise.

Entry	Substrate	R	Product	t[h]	Yield $[\%]$	E:Z
-1	36 a	Н	37a	20	$69^{[a]}$	0:100
2	36 b	TBS	37 b		91	40:60
3	36 c	Мe	37c	1.5	93	66:34
$\overline{4}$	36 d	MOM	37 d	ς	$Q1^{[a]}$	68:32

[a] Using 10 mol% of the catalyst.

It should be pointed out that the pronounced stereochemical preferences observed during these RCM reactions cannot be correlated with the thermodynamic stabilities of the isomeric products formed.[38d] Moreover, it is most unlikely that a direct interaction between the unprotected OH group of substrate 36a with the ruthenium catalyst (e.g. via phenolate complex formation)[50] accounted for the observed (Z) -selectivity in this particular case. Thus, cyclization of the methoxy protected diene $36c$ in the presence of one equivalent of salicylic acid methyl ester takes the same course as the cyclization of this substrate in the absence of the phenol in terms of yield and stereoselectivity. Since we are presently not able to provide a conclusive explanation for the observed stereochemical results, we take this case as an excellent opportunity to initiate in-depth theoretical investigations. The results of these studies will be reported in due course.

For the sake of the synthesis, it was fortunate to find that the methyl ether derivative 36c afforded macrolide 37c in excellent yield and reasonable selectivity in favor of the desired (E) -alkene. Substrate 36 c was particularly well accessible by direct esterification of fragment 35 with acid 12 (prepared in only one step as described above). Moreover, the retention times of (E) -37c and (Z) -37c were sufficiently different to allow separation of these isomers by conventional flash chromatography. (E) -37 c constituted the fully functional core of salicylihalamide and was converted into the target as outlined below.

Synthesis of (Z,Z)-2,4-heptadienoic acid amide and model studies on the copper-catalyzed enamide formation: Encouraged by a recent report on enamide formations by copper catalyzed cross-coupling reactions of vinyl iodides with carboxylic acid amides, $^{[15]}$ we envisaged to use this strategy for the final assembly of salicylihalamide. This publication, however, left open whether this method would be compatible with ester and unprotected OH groups; this aspect is not obvious as rather high loadings of the copper(i)thiophenecarboxylate (CuTC)^[41] catalyst and excess base are required. Moreover, it was not clear if a fairly labile (Z, Z) -configured $\alpha, \beta, \gamma, \delta$ -unsaturated amide can be used without loss of configurational integrity.[42] These issues were addressed by the model studies summarized below.

The required amide 42 was prepared as outlined in Scheme 7. Addition of iodide to methyl propynoic acid afforded (Z) -39 in good yield according to a literature procedure.[43] A subsequent Negishi coupling with butynylzinc chloride in the presence of catalytic amounts of $[Pd(PPh₃)₄]$ furnished product 40,^[44] which did undergo Lindlar reduction to afford the (Z,Z) -configurated ester 41. This step required

chloride, cat. $[Pd(PPh₃)₄]$, THF, 3 h, 90%; $[c]H₂$ (1 atm), Lindlar catalyst, cat. quinoline, CH_2Cl_2 , 50 min, 80%; [d] NH₄OH, NH₄Cl, 4 d, 62%.

careful monitoring by TLC to avoid overreduction. Aminolysis of ester 41 then provided the required amide 42 which was immediately used for the cross-coupling experiments.

We were pleased to see that this amide did undergo a smooth reaction with vinyl iodide 43a bearing a methoxy substituent on the arene ring in the presence of CuTC (50 mol%) and excess Rb_2CO_3 as the base in anhydrous dimethylacetamide (DMA) at 90 °C (Scheme 8). Note, however, that an excess of 42 was necessary to drive the

Scheme 8. [a] amide $\,$ **42** $\,$ (5 equiv), copper(i) thiophene-2-carboxylate (50 mol\%) , Rb₂CO₃ (3 equiv), DMA, 90 °C, 2 h, 89 % (R = Me, E: $Z =$ 2:1)), 86% $(R = H, only E)$; [b] benzamide, copper(i) thiophene-2-carboxylate (30 mol\%) , Cs_2CO_3 (1.5 equiv) , NMP, $90^{\circ}C$, 70% . NMP = Nmethylpyrrolidone, DMA = dimethyl acetamide.

conversion. The cross-coupling of substrate 43 b containing an unprotected phenol group proceeded similarly well. In the latter case, the resulting enamide 44 b was obtained as a single diastereoisomer in high yield. This result proved that the (Z) configuration of both alkene groups in the amide part was fully retained, whereas isomerization of the double bond of vinyl iodide **43b** $(E:\mathbb{Z} = 8:1)$ obviously could not be avoided. A third model reaction probing the stability of an ester group under the reaction condition also gave a satisfactory result $(45 \rightarrow 46)$.

Completion of the total synthesis: Since these model studies provided an encouraging outlook on the end game of the total synthesis of 1, the macrolide core 37 prepared by RCM was converted into a suitable vinyl iodide as shown in Scheme 9. For this purpose, the -OPMB group was cleaved off by means of DDQ,[45] the resulting primary alcohol 47 was oxidized with Dess – Martin periodinane,^[46] and aldehyde **48** thus formed was subjected to a Takai olefination.^[47] Specifically, treatment of 48 with CHI₃ and CrCl₂ in a mixed solvent system (THF/ 1,4-dioxane $1:6$ ^[48] furnished the desired vinyl iodide 49 as an inseparable mixture of isomers $(E:\mathbb{Z} = 9:1)$, as determined by NMR). This material was fully deprotected on exposure to $BBr₃$, since cleavage of the -OMe and the -OMOM group after the installation of the enamide might have endangered the valuable product (c.f. retrosynthetic analyis).

With the iodide 50 at hand, the stage was set for the final assembly of the target by the copper-catalyzed cross coupling technique. Gratifyingly, reaction of 50 with amide 42 (3 equiv) in the presence of CuTC (50 mol%) and Rb_2CO_3 (3 equiv) proceeded cleanly, affording a mixture of salicylihalamide A

Scheme 9. [a] DDQ, H_2O/CH_2Cl_2 1:18, RT, 12 h, 94%; [b] Dess-Martin periodinane, CH₂Cl₂, 0° C \rightarrow RT, 18 h, 87%; [c] CHI₂, CrCl₂, THF/1.4dioxane 1:6, RT, 16 h, 87% $(E:\mathbb{Z}=9:1)$; [d] BBr₃, CH₂Cl₂, $-78^{\circ}\text{C} \rightarrow \text{RT}$, 1 h, 88%; [e] amide **42** (3 equiv), copper(i) thiophene-2-carboxylate (50 mol\%) , Rb_2CO_3 (3 equiv), DMA, 90 °C, 2 h, 57%. DDQ = 2,3dichloro-5,6-dicyano-1,4-benzoquinone.

and B in 57% isolated yield $(1a:1b:ca. 2.5:1)$. The isomers could be separated by HPLC and were identical in all respects to the natural products.^[1, 9, 12] It is worth mentioning, however, that the copper-catalyzed enamide formation leading to 1 proceeded if Rb_2CO_3 was employed as the base. Although we can presently provide no satisfactory explanation, the use of $Cs₂CO₃$ failed to afford any of the desired target molecule under otherwise identical conditions, despite the fact that this base had worked well in model studies.[15]

Conclusion

A concise and inherently flexible approach to the potent antitumor agent salicylihalamide 1 was outlined. This strongly cytotoxic agent was assembled from three well accessible fragments by relying on the power of transition metal catalyzed $C-C$ and $C-X$ bond forming reactions. Scheme 10 illustrates this aspect. Particularly noteworthy are the rigorously chemo- and stereoselective hydrogenation reactions

Scheme 10. Summary of the total synthesis of salicylihalamide 1.

which guarantee reagent-control over the configuration of the newly formed chiral centers, the high yielding closure of the macrocyclic ring by RCM, and the use of the yet largely unexplored copper-catalyzed cross-coupling technique allowing the direct attachment of amides to unsaturated fragments. Importantly, it has been shown that this reaction tolerates

FULL PAPER A. Fürstner et al.

sensitive functional groups and therefore holds promise for further applications to complex targets in the future.

Finally, this total synthesis highlights once again the excellent performance of "second-generation" ruthenium carbenes for olefin metathesis. At the same time, however, it revealed that we still lack proper insight into the stereodetermining step of this transformation, as the configuration of the newly formed olefin may depend on subtle factors that are difficult to rationalize and foresee. Therefore it is obvious that further in-depth studies on this powerful transformation are necessary. The accompanying paper in this issue reporting the total synthesis of another potent antitumor agent, that is epothilone A and C, will further illustrate this notion.[49] Extensions of our studies on metathesis in general as well as additional syntheses of bioactive target molecules are currently underway and will soon be disclosed.

Experimental Section

General: All reactions were carried out under Ar. The solvents used were purified by distillation over the drying agents indicated and were transferred under Ar: THF, Et₂O (Mg/anthracene), CH₂Cl₂ (P₄O₁₀), CH₃CN, Et3N (CaH2), MeOH (Mg), DMF, DMA (Desmodur, dibutyltin dilaurate), hexane, toluene (Na/K). Flash chromatography: Merck silica gel 60 (230 -400 mesh). NMR: Spectra were recorded on a DPX 300 or DMX 600 spectrometer (Bruker) in the solvents indicated; chemical shifts (δ) are given in ppm relative to TMS, coupling constants (J) in Hz. IR: Nicolet FT-7199 spectrometer, wavenumbers in cm^{-1} . MS (EI): Finnigan MAT 8200 (70 eV), HRMS: Finnigan MAT 95. Melting points: Gallenkamp melting point apparatus (uncorrected). Optical rotation: Perkin Elmer 343 at $\lambda =$ 589 nm (Na-D line). Elemental analyses: Kolbe, Mülheim/Ruhr. All commercially available compounds (Lancaster, Aldrich) were used as received.

Starting materials and model studies

2-Allyl-6-hydroxy-benzoic acid (10) : A solution of BCl₂ $(1 \text{M} \text{ in } CH_2Cl_2)$ 30 mL, 30 mmol) was slowly added to a solution of compound 9 (1.00 g, 4.54 mmol)^[20] in CH₂Cl₂ (80 mL) at 0 °C and the resulting mixture was stirred for 5 h at ambient temperature. The organic phase was diluted with ethyl acetate (50 mL), washed with brine and dried over $Na₂SO₄$. Evaporation of the solvent and flash chromatography (hexanes/ethyl acetate/acetic acid 10:1:1) afforded acid 10 as colorless crystals (0.78 g, 96%). M.p. 98 – 99 °C; ¹H NMR (CDCl₃, 300 MHz): δ = 11.15 – 10.75 (br s, $1\,\mathrm{H}$), 7.38 (dd, $J = 8.2$, 7.7 Hz, $1\,\mathrm{H}$), 6.90 (dd, $J = 8.2$, 0.9 Hz, $1\,\mathrm{H}$), 6.79 (dd, $J = 7.7, 0.9$ Hz, 1H), 6.02 (ddt, $J = 17.0, 10.2, 6.3$ Hz, 1H), 5.07 – 4.96 (m, 2H), 3.76 (d, $J=6.3$ Hz, 2H); ¹³C NMR (CDCl₃, 75.5 MHz): $\delta = 175.2$, $163.6, 144.3, 137.3, 135.6, 122.7, 116.5, 115.7, 110.6, 40.1; IR (KBr): $\nu = 3047$,$ 2853, 2704, 2589, 1643, 1606, 1576, 1441, 1410, 1309, 1293, 1275, 1237, 1193, 1169, 1124, 1068, 1014, 1002, 915, 814, 792, 757, 707, 573 cm⁻¹; MS (EI): m/z (%): 178 (33) [M] , 160 (100), 132 (24), 115 (3), 104 (26), 77 (12), 63 (4), 51 (11); HR-MS (EI): $(C_{10}H_{10}O_3)$ calcd 178.0630; found 178.0632; elemental analysis calcd (%) for $C_{10}H_{10}O_3$ (178.20): C 67.41, H 5.66; found C 67.53, H 5.75.

(3S,4R)-3-(tert-Butyldimethylsilanyloxy)-4,7-dimethyl-oct-6-enoic methyl ester (20): 2,6-Lutidine (303 μ L, 2.60 mmol) and TBSOTf (360 μ L, 1.56 mmol) were added at 0° C to a solution of alcohol 19 (260 mg, 1.3 mmol)^[10] in CH₂Cl₂ (20 mL). After stirring for 1 h, the reaction was quenched with aq. NaOH $(2_N, 10_{mL})$, the mixture was extracted with CH₂Cl₂ (80 mL), the organic layer was washed with aq. sat. NH₄Cl (2 \times 10 mL) and brine (10 mL), dried over Na_2SO_4 and evaporated. Flash chromatography of the crude product (hexanes/ethyl acetate 20:1) afforded product 20 as a colorless syrup (389 mg, 95%). $[a]_D^{20} = -22.8$ ($c = 1.34$, CH_2Cl_2); ¹H NMR (CD₂Cl₂, 300 MHz): $\delta = 5.18 - 5.10$ (m, 1H), 4.16 - 4.08 $(m, 1H)$, 3.63 (s, 3H), 2.36 (d, J = 6.6 Hz, 2H), 2.04 – 1.92 (m, 1H), 1.83 – 1.72 (m, 1 H), $1.71 - 1.62$ (m, 1 H), 1.70 (s, 3 H), 1.60 (s, 3 H), 0.87 (s, 9 H), 0.86 (d, $J = 6.7$ Hz, 3H), 0.06 (s, 3H), 0.02 (s, 3H); ¹³C NMR (CD₂Cl₂, 75.5 MHz : $\delta = 173.0, 132.8, 123.3, 73.2, 51.7, 40.4, 38.5, 31.6, 25.9, 25.8, 18.3,$

17.9, 14.3, -4.5 , -4.7 ; IR (film): $\nu = 2956$, 2929, 2894, 2857, 1744, 1473, 1463, 1437, 1408, 1378, 1362, 1341, 1290, 1255, 1193, 1172, 1134, 1078, 1034, 1006, 982, 947, 886, 835, 812, 776, 665 cm⁻¹; MS (EI): m/z (%): 314 (<1) [M] , 299 (3), 283 (2), 257 (100), 225 (8), 199 (2), 182 (17), 147 (9), 131 (10), 115 (7), 109 (18), 89 (30), 73 (23), 69 (53), 41 (16); HR-MS (CI): $(C_{17}H_{34}O_3Si + H)$ calcd 315.2355; found 315.2358; elemental analysis calcd for $C_{17}H_{34}O_3Si$ (314.54): C 64.92, H 10.90; found C 65.04, H 11.03.

(3S,4R)-3-(tert-Butyldimethylsilanyloxy)-4,7-dimethyl-oct-6-en-1-ol (22 a): LiBEt₃H (1_M in THF, 1.62 mL, 1.62 mmol) was added at 0° C to a solution of ester 20 (255 mg, 0.81 mmol) in THF (20 mL). After stirring for 2 h at ambient temperature, the reaction was quenched with water, the aqueous layer was extracted with Et₂O (3×20 mL), the combined organic phases were dried over $Na₂SO₄$ and evaporated. Flash chromatography of the residue (hexanes/ethyl acetate 1:1) afforded product 22 a as a colorless syrup (193 mg, 83%). $\lbrack a \rbrack_{D}^{20} = -20.9$ ($c = 0.91$, CH₂Cl₂); ¹H NMR (CD₂Cl₂, 300 MHz : $\delta = 5.17 - 5.09 \text{ (m, 1H)}$, $3.78 \text{ (dt, } J = 6.8, 4.7 \text{ Hz, } 1 \text{ H})$, $3.69 \text{ (t, } J =$ 6.2 Hz, 2 H), $2.06 - 1.94$ (m, 1 H), $1.83 - 1.58$ (m, 11 H), 0.90 (s, 9 H), 0.86 (d, $J = 6.7$ Hz, 3H), 0.08 (s, 3H), 0.07 (s, 3H); ¹³C NMR (CD₂Cl₂, 75.5 MHz): δ = 132.5, 123.6, 74.7, 61.0, 40.0, 34.2, 32.0, 26.0, 25.8, 18.3, 17.9, 14.1, -4.3, -4.6 ; IR (KBr): $\nu = 3364$, 2958, 2929, 2885, 2857, 1472, 1463, 1407, 1377, 1361, 1256, 1083, 1062, 1034, 1005, 967, 938, 836, 774, 735, 666 cm⁻¹; MS (EI): m/z (%): 286 (<1) [M]⁺, 229 (18), 189 (6), 173 (3), 154 (11), 137 (26), 105 (11), 95 (24), 89 (13), 81 (40), 75 (38), 69 (100), 55 (10), 41 (20); HR-MS (CI): $(C_{16}H_{34}O_2Si + H)$ calcd 287.2406; found 287.2406; elemental analysis calcd for $C_{16}H_{34}O_2Si$ (286.53): C 67.07, H 11.96; found C 67.14, H 11.99.

 $(3S, 4R)$ -3-Methoxymethyl-4,7-dimethyl-oct-6-en-1-ol $(22b)$: Prepared in 98% yield as described above starting from ester 21 (366 mg, 1.5 mmol).[10] ¹H NMR (CD₂Cl₂, 300 MHz): δ = 5.17 – 5.09 (m, 1H), 4.66 (d, J = 6.8 Hz, $1\,\text{H}$), $4.60\,\text{(d, } J = 6.8\,\text{Hz, } 1\,\text{H}$), $3.98 - 3.52\,\text{(m, 3H), } 3.34\,\text{(s, } 3\,\text{H})$, $2.50 - 1.62\,\text{Hz}$ $(m, 6H)$, 1.68 (s, 3H), 1.60 (s, 3H), 0.86 (d, $J=6.4$ Hz, 3H); ¹³C NMR $(CD_2Cl_2, 75.5 MHz)$: $\delta = 132.5, 123.5, 96.5, 78.7, 60.7, 55.8, 37.4, 32.5, 31.8,$ $25.9, 17.9, 14.5; \text{IR (film): } \nu = 3436, 2962, 2931, 2886, 1672, 1460, 1405, 1377,$ 1343, 1218, 1151, 1100, 1038, 918, 860, 825, 764 cm⁻¹; MS (EI): *m*/z (%): 216 (-1) [M] , 184 (14), 154 (11), 139 (6), 121 (10), 109 (13), 97 (22), 87 (22), 69 (39) , 55 (24), 45 (100) 41 (27); HR-MS (CI): $(C_{12}H_{24}O_3 + H)$ calcd 217.1804; found 217.1803; elemental analysis calcd for $C_{12}H_{24}O_3$ (216.32): C 66.63, H 11.18; found C 66.72, H 11.16.

(3S,4R)-2-Allyl-6-hydroxybenzoic acid 3-(tert-butyldimethylsilanyloxy)- 4,7-dimethyloct-6-enyl ester (25a): A solution of alcohol 22a (175 mg, 0.66 mmol) and PPh₃ (194 mg, 0.74 mmol) in Et₂O (25 mL) was added dropwise to a suspension of acid 10 (132 mg, 0.74 mmol) and DEAD (116 μ L, 0.74 mmol) in Et₂O (35 mL). After stirring for 20 h, the suspension was concentrated to a total volume of ca. 10 mL, the precipitates were filtered off, the filtrate was dried over $Na₂SO₄$ and evaporated, and the residue was purified by flash chromatography (hexanes/ethyl acetate 30:1) to afford ester **25 a** as a colorless syrup (265 mg, 96%). $[\alpha]_D^{20} = -26.4$ (*c* = 1.10, CH₂Cl₂); ¹H NMR (CD₂Cl₂, 300 MHz): $\delta = 11.18$ (s, 1H), 7.33 (dd, J = $8.3, 7.5$ Hz, 1 H), 6.85 (dd, $J = 8.3, 0.7$ Hz, 1 H), 6.76 (dd, $J = 7.5, 0.7$ Hz, 1 H), 6.01 (ddt, $J = 16.6$, 10.2, 6.2 Hz, 1H), 5.18 – 5.09 (m, 1H), 5.03 (dd, $J = 10.2$, 1.6 Hz, 1 H), 4.97 (dd, $J = 16.6$, 1.7 Hz, 1 H), $4.53 - 4.38$ (m, 2 H), $3.84 - 3.76$ $($, m, 1H), 3.72 (dd, $J = 6.1$, 1.3 Hz, 2H), 2.07 – 1.65 (m, 5H), 1.67 (s, 3H), 1.59 (s, 3H), 0.91 (s, 9H), 0.88 (d, J = 6.8 Hz, 3H), 0.07 (s, 3H), 0.06 (s, 3H); ¹³C NMR (CD₂Cl₂, 75.5 MHz): δ = 171.8, 163.1, 143.3, 138.1, 134.6, 132.7, 123.4, 122.7, 116.4, 115.6, 112.7, 72.6, 64.1, 40.6, 40.1, 31.8, 31.0, 26.0, 25.8, $18.3, 17.9, 14.0, -4.3, -4.5; \text{IR (film): } v = 3079, 3060, 2958, 2929, 2886, 2857,$ 1730, 1662, 1608, 1579, 1451, 1409, 1377, 1342, 1312, 1296, 1219, 1194, 1165, 1119, 1097, 1067, 1006, 912, 867, 836, 817, 774, 713, 667 cm⁻¹; MS (EI): *m*/z $(\%)$: 446 (<1) $[M]^+$, 389 (5), 314 (4), 269 (2), 235 (12), 171 (5), 161 (100), 133 (9), 69 (12), 41 (6); HR-MS (CI): $(C_{26}H_{42}O_4Si + H)$ calcd 447.2931; found 447.2930; elemental analysis calcd for $C_{26}H_{42}O_4Si$ (446.70): C 69.91, H 9.48; found C 70.18, H 9.12.

(3S,4R)-2-Allyl-6-hydroxybenzoic acid 3-methoxymethyl-4,7-dimethyloct-6-enyl ester (25 b): Prepared as described above from alcohol 22 b (382 mg, 1.77 mmol) and acid 10 (315 mg, 2.12 mmol). Colorless syrup (537 mg, 81 %). $[\alpha]_D^{20} = -39.0$ ($c = 1.85$, CH₂Cl₂); ¹H NMR (CD₂Cl₂, 300 MHz): $\delta =$ 11.09 (s, 1H), 7.33 (dd, $J = 8.3$, 7.5 Hz, 1H), 6.85 (dd, $J = 8.3$, 1.0 Hz, 1H), 6.76 (dd, $J = 7.5$, 0.9 Hz, 1H), 6.01 (ddt, $J = 16.4$, 10.2, 6.1 Hz, 1H), 5.17 – $4.93 \text{ (m, 3H)}, 4.67 \text{ (d, } J = 6.8 \text{ Hz}, 1 \text{ H}), 4.60 \text{ (d, } J = 6.8 \text{ Hz}, 1 \text{ H}), 4.58 - 4.42 \text{ K}$ $(m, 2H)$, 3.72 (d, 6.1 Hz, 2H), 3.67 – 3.58 $(m, 1H)$, 3.36 (s, 3H), 2.08 – 1.81 $(m, 5H)$, 1.68 (s, 3H), 1.59 (s, 3H), 0.89 (d, $J=6.4$ Hz, 3H); ¹³C NMR $(CD_2Cl_2, 75.5 MHz)$: $\delta = 171.6, 162.9, 143.3, 138.2, 134.6, 132.9, 123.1, 122.7,$

116.3, 115.5, 112.8, 96.6, 78.7, 63.8, 56.0, 40.5, 37.2, 31.7, 29.4, 25.8, 17.9, 14.2; IR (film): $\nu = 3385, 3077, 2965, 2931, 2889, 2823, 1727, 1661, 1607, 1578, 1451,$ 1377, 1343, 1313, 1296, 1249, 1220, 1165, 1142, 1121, 1100, 1038, 987, 916, 817, 769, 713 cm⁻¹; MS (EI): m/z (%): 376 (<1) [M]⁺, 344 (15), 205 (13), 183 (13), 160 (100), 133 (10), 121 (24), 93 (8), 69 (22), 55 (7), 45 (37); HR-MS (CI): $(C_2H_{32}O_5 + H)$ calcd 377.2328; found 377.2325; elemental analysis calcd for $C_{22}H_{32}O_5$ (376.49): C 70.19, H 8.57; found C 70.28, H 8.52.

(3S,4R)-2-Allyl-6-(tert-butyldimethylsilanyloxy)-benzoic acid 3-methoxymethyl-4,7-dimethyloct-6-enyl ester (25 c): A solution of compound 25 b (113 mg, 0.3 mmol), imidazole (204 mg, 3 mmol) and TBSCl (362 mg, 2.4 mmol) in DMF (20 mL) was stirred for 18 h at ambient temperature. The mixture was diluted with EtOAc (50 mL) and washed with brine (3 \times 20 mL). Drying of the organic phase over $Na₂SO₄$, evaporation of the solvent followed by flash chromatography of the residue (hexanes/ethyl acetate 20:1) afforded product $25c$ as a colorless syrup (145 mg, 99%). $[\alpha]_{\text{D}}^{20}$ = -21.3 (c = 1.47, CH₂Cl₂); ¹H NMR (CD₂Cl₂, 300 MHz): δ = 7.20 (dd, $J = 8.2, 7.6 \text{ Hz}, 1 \text{ H}$), 6.82 (dd, $J = 7.6 \text{ Hz}, 1 \text{ H}$), 6.73 (dd, $J = 8.2 \text{ Hz}, 1 \text{ H}$), 5.92 $(ddt, J=16.2, 9.5, 6.6 \text{ Hz}, 1 \text{ H}), 5.14-5.01 \text{ (m, 3H)}, 4.65 \text{ (d, } J=6.8 \text{ Hz}, 1 \text{ H}),$ 4.60 (d, $J = 6.8$ Hz, 1 H), $4.47 - 4.37$ (m, 1 H), $4.33 - 4.24$ (m, 1 H), $3.57 - 3.52$ $(m, 1H)$, 3.36 (s, 3H), 3.32 (d, 6.6 Hz, 2H), 2.07 - 1.94 $(m, 1H)$, 1.90 - 1.77 $(m, 4H), 1.67$ (s, 3H), 1.57 (s, 3H), 0.97 (s, 9H), 0.88 (d, $J = 6.4$ Hz, 3H), 0.23 (s, 6H); ¹³C NMR (CD₂Cl₂, 75.5 MHz): δ = 168.3, 152.7, 138.9, 137.0, 132.8, 130.3, 127.2, 123.2, 122.2, 117.2, 116.3, 96.6, 78.9, 63.1, 55.9, 37.9, 37.4, $31.7, 29.7, 25.8, 25.7, 18.3, 17.9, 14.4, -4.3; \text{ IR (film): } v = 3078, 2962, 2931,$ 2887, 2859, 2823, 1731, 1662, 1640, 1594, 1584, 1464, 1409, 1377, 1363, 1285, 1264, 1211, 1142, 1108, 1064, 1039, 994, 969, 917, 841, 806, 783, 741, 719, 669, 578, 555 cm⁻¹; MS (EI): m/z (%): 490 (< 1) [M]⁺, 433 (16), 401 (7), 275 (44), 249 (37), 235 (100), 199 (27), 167 (17), 137 (19), 81 (21), 69 (41), 57 (8), 45 (57); HR-MS (CI): $(C_{28}H_{46}O_5Si + H)$ calcd 491.3193; found 491.3193; elemental analysis calcd for $C_{28}H_{46}O_5Si$ (490.76): C 68.53, H 9.45; found C 68.62, H 9.54.

Representative procedure for RCM

(Z)-(9S,10R)-9-(tert-Butyldimethylsilanyloxy)-4-hydroxy-10-methyl-7,8,9,10,11,14-hexahydro-6-oxa-benzocyclododecen-5-one [(Z)-26 a]: Complex 24 (4.2 mg, 0.005 mmol) was added to a solution of diene 25 a (44.6 mg, 0.1 mmol) in toluene (50 mL). After stirring for 3 h at 80° C, the reaction was quenched with ethyl vinyl ether (1 mL). Evaporation of the solvent followed by flash chromatography (hexanes/ethyl acetate 50:1) of the crude product afforded cycloalkene (Z)-26 a as a colorless syrup (34.7 mg, 89%). $[\alpha]_{\text{D}}^{20}$ = -63.7 (c = 1.02, CH₂Cl₂); ¹H NMR (C₆D₆, 300 MHz): δ = 12.12 (s, $1\,\text{H}$), $7.02 - 6.92 \,(\text{m}, 2\,\text{H})$, $6.48 \,(\text{dd}, J = 6.6, 2.1 \, \text{Hz}, 1\,\text{H})$, $5.32 - 5.13 \,(\text{m}, 2\,\text{H})$, $4.42 - 4.32$ (m, 1H), 4.17 (dd, $J = 15.0$, 9.3 Hz, 1H), 3.85 (dt, $J = 10.1$, 3.2 Hz, $1\,\mathrm{H}$), $3.81 - 3.75$ (m, $1\,\mathrm{H}$), 2.85 (dd, $J = 15.0$, $2.4\,\mathrm{Hz}$, $1\,\mathrm{H}$), $2.08 - 1.92$ (m, $1\,\mathrm{H}$), $1.70 - 1.52$ (m, 3H), $1.35 - 1.24$ (m, 1H), 0.94 (s, 9H), 0.86 (d, $J = 6.7$ Hz, 3H), 0.00 (s, 3H), -0.01 (s, 3H); ¹³C NMR (C₆D₆, 75.5 MHz): δ = 172.1, 164.5, 144.8, 134.8, 131.2, 127.8, 123.1, 116.8., 111.9, 71.7, 64.8, 39.8, 35.4, $31.8, 31.3, 26.0, 18.1, 13.6, -4.4; \text{IR (KBr): } v = 3425, 3057, 3022, 3006, 2957,$ 2930, 2895, 2857, 1658, 1631, 1607, 1574, 1471, 1448, 1386, 1347, 1311, 1289, 1251, 1216, 1158, 1127, 1089, 1071, 1052, 1017, 1005, 965, 932, 904, 876, 862, 836, 819, 791, 775, 714, 699, 662, 592 cm⁻¹; MS (EI): m/z (%): 390 (<1) [M] , 375 (1), 333 (100), 315 (6), 285 (9), 241 (10), 205 (12), 171 (14), 161 (15), 149 (12), 131 (16), 93 (28), 73 (34), 59 (7), 41 (9); HR-MS (CI): $(C₂₂H₃₄O₄Si + H)$ calcd 391.2305; found 391.2307; elemental analysis calcd for C22H34O4Si (390.60): C 67.65, H 8.77; found C 67.57, H 8.84.

The following products were obtained analogously:

(Z)-(9S,10R)-4-Hydroxy-9-methoxymethyl-10-methyl-7,8,9,10,11,14-hexa**hydro-6-oxa-benzocyclododecen-5-one [(Z)-26b]**: Colorless syrup. $[\alpha]_D^{20}$ = -82.5 (c = 0.89, CH₂Cl₂); ¹H NMR (C₆D₆, 300 MHz): δ = 12.23 (s, 1H), $7.01 - 6.93$ (m, 2H), 6.46 (dd, $J = 6.3$, 2.5 Hz, 1H), 5.26 – 5.07 (m, 2H), 4.46 $(d, J=6.9 \text{ Hz}, 1 \text{ H}), 4.37 (d, J=6.9 \text{ Hz}, 1 \text{ H}), 4.38-4.28 \text{ (m, 1 H)}, 4.08 \text{ (dd,$ $J = 14.7, 8.9$ Hz, 1H), 3.85 (dt, $J = 10.9, 3.1$ Hz, 1H), 3.66 – 3.60 (m, 1H), 3.11 (s, 3H), $2.89 - 2.80$ (m, 1H), $2.04 - 1.49$ (m, 4H), $1.41 - 1.30$ (m, 1H), 0.87 (d, J = 6.7 Hz, 3H); ¹³C NMR (C₆D₆, 75.5 MHz): δ = 172.1, 164.5, 144.8, 134.8, 131.3, 127.6, 123.1, 116.8., 112.0, 95.9, 77.8, 65.2, 55.0, 36.6, 35.5, $31.9, 28.9, 13.7; \text{IR (KBr): } v = 3419, 3051, 2961, 2934, 1750, 1716, 1659, 1606,$ 1575, 1450, 1385, 1312, 1295, 1250, 1220, 1166, 1152, 1125, 1100, 1068, 1038, 916, 818, 775, 738, 711, 542 cm⁻¹; MS (EI): m/z (%): 320 (16) [M]⁺, 288 (7), 258 (29), 240 (3), 214 (5), 201 (14), 172 (22), 160 (13), 134 (8), 115 (8), 99 (4), 85 (44), 81 (6), 69 (1), 55 (12), 45 (100); HR-MS (EI): $(C_{18}H_{24}O_5)$ calcd 320.1624; found 320.1625; elemental analysis calcd for $C_{18}H_{24}O_5$ (320.39): C 67.48, H 7.55; found C 67.35, H 7.62.

(9S,10R)-4-(tert-Butyldimethylsilanyloxy)-9-methoxymethyl-10-methyl-

7,8,9,10,11,14-hexahydro-6-oxa-benzocyclododecen-5-one (26 c): Colorless syrup. $E:Z = 30:70$; $[\alpha]_D^{20} = -10.4$ $(c = 0.73, \text{ CH}_2\text{Cl}_2)$; data of (E) -26 c: ¹H NMR (C₆D₆, 300 MHz): δ = 6.94 (dd, J = 8.2, 7.6 Hz, 1 H), 6.68 – 6.53 (m, 2H), 5.45 - 4.96 (m, 2H), 4.62 (s, 2H), 4.40 - 3.57 (m, 4H), 3.32 - 3.23 (m, 1 H), 3.21 (s, 3 H), 2.21 – 1.48 (m, 5 H), 0.98 (s, 9 H), 0.87 (d, $J = 6.5$ Hz, 3 H), 0.13 (s, 6H); ¹³C NMR (C₆D₆, 75.5 MHz): δ = 168.5, 153.2, 140.4, 130.8, 129.9, 128.3, 128.0, 123.2, 117.4, 96.4, 76.3, 63.0, 55.2, 38.3, 37.8, 34.4, 31.9, $25.7, 18.2, 14.0, -4.4.$

Data of (Z)-26 c: ¹H NMR (C₆D₆, 300 MHz): δ = 6.97 (dd, J = 8.2, 7.6 Hz, $1\,\mathrm{H}$), 6.68 – 6.53 (m, 2 H), 5.45 – 4.96 (m, 2 H), 4.50 (d, $J = 6.8$ Hz, 1 H), 4.44 $(d, J=6.8 \text{ Hz}, 1 \text{ H}), 4.40-3.57 \text{ (m, 4H)}, 2.96-2.88 \text{ (m, 1H)}, 3.12 \text{ (s, 3H)},$ $2.21 - 1.48$ (m, 5H), 0.99 (s, 9H), 0.84 (d, $J = 6.5$ Hz, 3H), 0.10 (s, 3H), 0.09 $(s, 3H);$ ¹³C NMR (C₆D₆, 75.5 MHz): δ = 168.1, 152.5, 139.8, 130.1, 129.6, 128.4, 128.3, 122.6, 116.7, 96.3, 78.9, 62.3, 55.1, 36.3, 32.1, 30.5, 29.5, 25.7, 18.2, 13.8, -4.4; IR (KBr): $\nu = 3068$, 2957, 2931, 2887, 2858, 2822, 1729, 1660, 1592, 1583, 1464, 1378, 1363, 1284, 1261, 1211, 1153, 1106, 1065, 1040, 974, 939, 919, 902, 842, 783, 739, 701, 670, 574 cm⁻¹; MS (EI): *m*/z (%): 434 (-1) [M] , 377 (75), 315 (19), 247 (22), 221 (24), 207 (42), 141 (10), 73 (32), 57 (9), 45 (100); HR-MS (CI): $(C_{24}H_{38}O_5Si + H)$ calcd 435.2567; found 435.2562; elemental analysis calcd for C₂₄H₃₈O₅Si (434.65): C 66.32, H 8.81; found C 66.28, H 8.88.

(Z)-(9R,10R)-9-(tert-Butyldimethylsilanyloxy)-4-hydroxy-10-methyl-

7,8,9,10,11,14-hexahydro-6-oxa-benzocyclododecen-5-one [(Z)-28]: Colorless syrup. $[\alpha]_D^{20} = -54.2$ (c=0.99, CH₂Cl₂); ¹H NMR (C₆D₆, 600 MHz): δ = 12.32 (s, 1H), 6.99 (dd, J = 8.3, 7.2 Hz, 1H), 6.95 (dd, J = 8.3, 1.6 Hz, 1 H), 6.50 (dd, $J = 7.2$, 1.6 Hz, 1 H), 5.52 – 5.46 (m, 1 H), 5.35 – 5.30 (m, 1 H), $4.49 - 4.42$ (m, 1H), $4.14 - 4.05$ (m, 2H), $3.32 - 3.27$ (m, 1H), $3.26 - 3.19$ (m, $1\,\mathrm{H}$), $2.37 - 2.31\,\mathrm{(m, 1\,H)}$, $1.98 - 1.93\,\mathrm{(m, 1\,H)}$, $1.59 - 1.56\,\mathrm{(m, 1\,H)}$, $1.54 - 1.45$ (m, 2H), 0.98–0.94 (m, 3H), 0.92 (s, 9H), -0.01 (s, 3H), -0.06 (s, 3H);
¹³C NMR (C₆D₆, 151 MHz): δ = 172.4, 164.8, 144.2, 134.7, 130.7, 126.5, 122.9, 116.7, 111.9, 74.4, 62.6, 40.1, 34.6, 33.6, 30.7, 26.0, 18.2, 17.5, 4.2, -4.5 ; IR (KBr): $v = 3450$, 3009, 2956, 2930, 2895, 2857, 1734, 1655, 1605, 1576, 1471, 1463, 1449, 1389, 1360, 1337, 1309, 1295, 1251, 1217, 1177, 1166, 1117, 1073, 1037, 995, 905, 886, 862, 837, 816, 773, 721, 710, 669, 598 cm⁻¹; MS (EI): m/z (%): 390 (<1) [M]⁺, 375 (1), 333 (100), 315 (6), 285 (9), 241 (10), 205 (12), 171 (14), 161 (15), 149 (12), 131 (16), 93 (28), 73 (34), 59 (7), 41 (9).

Total synthesis of salicylihalamide

N-Acylsultam (31):^[39] A solution of *n*BuLi (1.6 *M* in hexane, 14.20 mL, 22.7 mmol) and cyclohexylisopropylamine (382 μ L, 2.27 mmol) was added over 60 min at -78° C to a solution of N-acylsultam (30) (6.17 g, 22.7 mmol) in THF (120 mL, 0.19 m). The resulting lithium enolate was stirred at -78 °C for 1 h before freshly distilled dimethylallyl bromide (7.94 mL, 68.2 mmol) in HMPA (11.92 mL, 68.2 mmol) was added. The reaction mixture was allowed to warm to ambient temperature, was quenched with water (150 mL) and extracted with Et₂O (3×150 mL). Drying (Na2SO4) of the combined organic phases and evaporation of the solvents gave the crude product which was crystallized from methanol to afford pure 31 (6.09 g, 79%) as a colorless solid. $[\alpha]_{D}^{20} = -52.5$ ($c = 2.2$, CH₂Cl₂); m.p. 87–89 °C; ¹H NMR (CD₂Cl₂, 300 MHz): δ = 5.09 (t, J = 7.5 Hz, 1H), 3.85 (dd, $J = 4.8$, 7.5 Hz, 1H), 3.43 (q, $J = 11.1$ Hz, 2H), 3.12 – 3.05 (m, 1H), 2.76 (d, $2H$), $2.44 - 2.34$ (m, $2H$), $2.13 - 1.82$ (m, $3H$), 1.63 (s, $3H$), 1.57 (s, $3H$), $1.42-1.15$ (m, $2H$), 1.11 (dd, $J=6.8$ Hz, $6H$), 0.96 (s, $3\,\text{H}$); ¹³C NMR (CD₂Cl₂, 75.5 MHz): δ = 176.5, 134.4, 121.5, 65.6, 54.7, 53.3, 48.7, 48.2, 45.4, 40.6, 34.5, 33.3, 26.9, 26.1, 21.1, 20.2, 18.0, 16.7; IR (KBr): - 2970, 2930, 2882, 1689, 1461, 1394, 1336, 1272, 1239, 1220, 1166, 1135, 1120, 1062, 1037, 976, 769 cm⁻¹; MS (EI): m/z (%): 339 (27) [M]⁺, 271 (57), 152 (16), 136 (11), 135 (78), 134 (27), 125 (10), 107 (12), 97 (70), 96 (100), 81 (15), 79 (10), 69 (40), 67 (11), 55 (61), 43 (11), 41 (37); HR-MS (EI): (C18H29NO3S) calcd 339.1868; found 339.1867.

(2S)-2,5-Dimethyl-hex-4-enoic acid (ent-15): Aqueous H_2O_2 (30% w/w, 22.7 mL) and a suspension of $LiOH H₂O$ (7.89 g, 188.3 mmol) in water (40 mL) were added at 0° C to a solution of compound 31 (15.98 g, 47.1 mmol) in THF (400 mL) and water (100 mL). The reaction was first stirred for 60 min at 0° C and then at ambient temperature for 24 h. Acidification with HCl (2 M , 500 mL), extraction with CH₂Cl₂ (3 \times 300 mL), drying of the combined organic layers over $Na₂SO₄$, evaporation of the

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solvents and trituration of the residue with pentane furnished the insoluble auxiliary. The soluble carboxylic acid was purified by column chromatography (pentane/Et₂O 4:1, 1% acetic acid) affording compound ent-15 (6.36 g, 95%) as a colorless liquid. $\lbrack a \rbrack_{D}^{20} = +7.4$ ($c = 2.08$, CH₂Cl₂); ¹H NMR $(CDCl_3$, 300 MHz): $\delta = 10.50 - 10.00$ (br s, 1H), 5.12 – 5.04 (m, 1H), 2.52 – 2.28 (m, 2H), 2.19 – 2.08 (m, 1H), 1.67 (s, 3H), 1.59 (s, 3H), 1.14 (d, $J =$ 6.9 Hz, 3H); ¹³C NMR (CDCl₃, 75.5 MHz): δ = 183.0, 134.0, 120.9, 39.8, 31.8, 25.8, 17.8, 16.3; IR (KBr): - 2975, 2935, 2661, 1710, 1463, 1417, 1378, 1338, 1287, 1245, 1226, 1185, 1125, 1083, 1049, 933, 856, 812, 778, 625 cm⁻¹; MS (EI): m/z (%): 142 (24) [M] , 124 (2), 109 (1), 97 (3), 87 (5), 81 (5), 74 (17), 69 (100), 55 (10), 41 (55); HR-MS (EI): $(C_8H_{14}O_2)$ calcd 142.0994; found 142.0993.

(2S)-2,5-Dimethylhex-4-enoyl chloride (*ent*-16): Chloroenamine 17 (7.30 g, 54.88 mmol $[25]$ was slowly added through a syringe to a solution of acid *ent*-15 (6.50 g, 45.78 mmol) in CH_2Cl_2 (80 mL). After stirring for 90 min, the solvent was evaporated in vacuo affording acid chloride ent-16 as a colorless oil. The crude product was dissolved in THF (80 mL) and was directly used in the next step.

(4S)-4,7-Dimethyl-3-oxo-oct-6-enoic methyl ester (ent-18): A solution of nBuLi (1.6 M in hexane, 109.0 mL, 174.4 mmol) was slowly added at -78° C to a solution of diisopropylamine (22.60 mL, 159.4 mmol) in THF (400 mL). The reaction mixture was stirred for 30 min at -20° C prior to the slow addition of methyl acetate (12.70 mL, 159.4 mmol) at -78° C. After stirring for 1 h, a solution of acid chloride ent-16 (45.78 mmol, crude) in THF (80 mL) was added and the reaction mixture was quickly warmed to ambient temperature. After 2 h the reaction was quenched by addition of aq. sat. NH₄Cl. Extraction with Et₂O, drying of the combined organic phases over $Na₂SO₄$, evaporation of the solvent and flash chromatography (hexanes/ethyl acetate 15:1) afforded keto ester ent-18 as a colorless oil $(7.31 \text{ g}, 81 \text{ %}).$ $[\alpha]_D^{20} = +35.1$ $(c = 1.54, CH_2Cl_2)$; according to NMR, the product consists of a 9:1 mixture of keto-enol tautomers. NMR Data for major form: ¹H NMR (CD₂Cl₂, 300 MHz): δ = 5.09 – 5.02 (m, 1H), 3.69 (s, 3H), 3.47 (s, 2H), 2.69 - 2.58 (m, 1H), 2.36 - 2.24 (m, 1H), 2.14 - 2.02 (m, 1 H), 1.69 (s, 3 H), 1.61 (s, 3 H), 1.07 (d, $J = 7.0$ Hz, 3 H); ¹³C NMR (CD₂Cl₂, 75.5 MHz : $\delta = 206.6, 168.1, 134.4, 121.2, 52.4, 48.1, 47.2, 31.5, 25.8, 17.9,$ 15.8; IR (KBr): $\nu = 2971, 2934, 1752, 1715, 1653, 1626, 1450, 1438, 1405,$ 1377, 1318, 1237, 1195, 1155, 1119, 1039, 1006, 849, 842, 805, 778, 739, 703, 658 cm⁻¹; MS (EI): m/z (%): 198 (18) [M]⁺, 180 (8), 166 (4), 143 (13), 130 (54), 125 (31), 109 (20), 101 (29), 96 (25), 81 (14), 74 (15), 69 (100), 55 (38), 41 (56); HR-MS (EI): (C₁₁H₁₈O₃) calcd 198.1256; found 198.1254.

(3R,4S)-3-Hydroxy-4,7-dimethyl-oct-6-enoic methyl ester (ent-19): A stainless steel autoclave (200 mL) was charged with a solution of compound ent-18 (2.80 g, 14.12 mmol) in MeOH (80 mL). After addition of $[(S)$ - $BINAP)RuCl₂$]₂ \cdot NEt₃ (3.1 mm in THF, 18.4 mL, 0.057 mmol, 0.4 mol%) the autoclave was pressurized with H_2 (4 atm) and the reaction mixture was stirred for 4 h at 80 °C. After venting the autoclave, the solvent was removed in vacuo and the residue was purified by flash chromatography (hexanes/ethyl acetate 6:1) affording product ent-19 (2.654 g, 94%) as a colorless oil. $[a]_D^{20} = +20.5$ $(c=1.29, CH_2Cl_2);$ ¹H NMR $(CD_2Cl_2,$ 300 MHz : $\delta = 5.14 \text{ (ddsept, } J = 7.8, 6.8, 1.4 \text{ Hz}, 1 \text{ H}$), 3.83 (dddd, $J = 9.5$, 6.1, 4.0, 2.6 Hz, 1 H), 3.68 (s, 3 H), 2.82 (d, $J = 4.0$ Hz, 1 H), 2.50 (dd, $J = 16.1$, 2.9 Hz, 1 H), 2.38 (dd, $J = 16.1$, 9.6 Hz, 1 H), 2.15 (ddd, $J = 14.2$, 6.8, 4.8 Hz, $1\,\mathrm{H}$), 1.89 (ddd, $J = 14.2, 8.6, 7.8$ Hz, $1\,\mathrm{H}$), 1.70 (q, 1.3 Hz, $3\,\mathrm{H}$), 1.61 (d, $J =$ $(0.8 \text{ Hz}, 3\text{ H}), 1.59 \text{ (dddq, } J = 8.4, 6.1, 4.8, 6.8 \text{ Hz}, 1\text{ H}), 0.86 \text{ (d, } J = 6.8 \text{ Hz},$ $3\,\text{H}$); ¹³C NMR (CD₂Cl₂, 75.5 MHz): δ = 174.1, 133.0, 123.0, 72.0, 52.0, 39.6, 38.5, 31.2, 25.9, 17.9, 15.3; IR (neat): $\nu = 3472, 2964, 2925, 2881, 1739, 1438,$ 1405, 1377, 1339, 1288, 1260, 1196, 1170, 1113, 1051, 1018, 990, 880, 846 cm⁻¹; MS (EI): m/z (%): 200 (11) [M]⁺, 182 (55), 167 (9), 150 (14), 122 (40), 109 (62), 107 (56), 103 (24), 93 (25), 81 (22), 69 (84), 55 (57), 41 (100), 29 (37); HR-MS (EI): $(C_{11}H_{20}O_3)$ calcd 200.1412; found 200.1413.

(3R,4S)-3-Methoxymethyl-4,7-dimethyl-oct-6-enoic methyl ester (ent-21): N,N-Dimethylaminopyridine (284 mg, 2.33 mmol), diisopropylethylamine (12.20 mL, 70.0 mmol) and MOMCl (5.31 mL, 70.0 mmol) were added to a solution of compound ent-19 (4.67 g, 23.3 mmol) in CH_2Cl_2 (500 mL). Stirring at ambient temperature was continued for 40 h. After dilution with ethyl acetate (600 mL), the organic phase was washed with brine ($3 \times$ 200 mL), dried over $Na₂SO₄$, evaporated, and the crude product was purified by flash chromatography (hexanes/ethyl acetate 10:1) affording *ent*-21 as a colorless syrup (5.11 g, 89%). $\lbrack a \rbrack_{D}^{20} = +15.6$ ($c = 1.90$, CH₂Cl₂); ¹H NMR (CD₂Cl₂, 300 MHz): δ = 5.15 – 5.08 (m, 1H), 4.61 (s, 2H), 3.95 – $3.87 \text{ (m, 1H)}, 3.65 \text{ (s, 3H)}, 3.30 \text{ (s, 3H)}, 2.44 \text{ (d, } J = 7.4 \text{ Hz}, 1 \text{ H}), 2.43 \text{ (d,)}$

4.9 Hz, 1H), $2.08 - 1.95$ (m, 1H), $1.88 - 1.75$ (m, 2H), 1.70 (s, 3H), 1.60 (s, 3H), 0.86 (d, $J = 6.5$ Hz, 3H); ¹³C NMR (CD₂Cl₂, 75.5 MHz): $\delta = 172.7$, 133.0, 123.0, 96.7, 78.8, 55.8, 51.8, 37.7, 36.7, 31.5, 25.9, 17.9, 14.5; IR (KBr): - 2962, 2932, 2889, 2824, 1742, 1673, 1437, 1378, 1343, 1290, 1272, 1214, 1194, 1173, 1150, 1100, 1043, 976, 919, 857, 821 cm⁻¹; MS (EI): *m*/z (%): 244 (-1) [M] , 212 (12), 194 (3), 182 (18), 167 (3), 155 (4), 139 (6), 121 (19), 103 (23), 81 (9), 69 (33), 55 (24), 45 (100), 41 (25), 29 (9); HR-MS (CI): $(C_{13}H_{24}O_3 + H)$ calcd 245.1753; found 245.1754.

(5R,6S)-5-Methoxymethyl-6,9-dimethyl-3-oxo-dec-8-enoic tert-butyl ester (32): A solution of tert-butyl acetate (8.37 mL, 62.40 mmol) in THF (20 mL) was slowly added to a solution of LiHMDS $(10.44 \text{ g}, 62.40 \text{ mmol})$ in THF (200 mL) at -45° C. The temperature was raised to -30° C over a period of 90 min. A solution of compound ent-21 (2.55 g, 10.43 mmol) in THF (40 mL) was then added at -40° C. Stirring was continued for 3 h while the temperature was increased to -30° C. Quenching of the reaction with aq. sat. NH₄Cl, extraction with Et₂O, drying of the combined organic phases over $Na₂SO₄$, evaporation of the solvent and flash chromatography (hexanes/ethyl acetate 10:1) afforded product 32 as a colorless syrup $(3.26 \text{ g}, 95\%)$. $[\alpha]_D^{20} = +18.1$ $(c=1.29, \text{ CH}_2\text{Cl}_2)$; according to NMR, the product consists of a 13:1 mixture of the keto-enol tautomers. NMR Data for the major keto form: ¹H NMR (CD₂Cl₂, 300 MHz): $\delta = 5.16 - 5.08$ (m, 1H), 4.58 (s, 2H), 4.01 - 3.94 (m, 1H), 3.37 (s, 2H), 3.29 (s, 3H), 2.73 (dd, $J = 16.2, 8.7$ Hz, 1H), 2.52 (dd, $J = 16.2, 3.2$ Hz, 1H), 2.04 – 1.92 (m, 1H), $1.87 - 1.76$ (m, 2H), 1.70 (s, 3H), 1.59 (s, 3H), 1.45 (s, 9H), 0.86 (d, $J =$ 6.4 Hz, 3H); ¹³C NMR (CD₂Cl₂, 75.5 MHz): δ = 202.5, 166.7, 133.0, 122.9, 96.7, 81.9, 77.7, 55.9, 51.9, 44.4, 37.6, 31.7, 28.1, 25.9, 18.0, 14.5; IR (neat): $\nu =$ 2975, 2931, 2824, 1738, 1717, 1643, 1456, 1408, 1393, 1369, 1319, 1286, 1253, 1212, 1150, 1099, 1040, 944, 919, 840 cm⁻¹; MS (EI): m/z (%): 328 (<1) [M] , 296 (2), 266 (1), 240 (25), 223 (5), 210 (15), 181 (9), 139 (15), 123 (14), 109 (22), 97 (15), 81 (10), 69 (32), 57 (72), 45 (100), 41 (31), 29 (13); HR-MS (CI): $(C_{18}H_{32}O_5 + H)$ calcd 329.2328; found 329.2328.

(3S,5R,6S)-3-Hydroxy-5-methoxymethyl-6,9-dimethyl-dec-8-enoic tert-butyl ester (33): A stainless steel autoclave (50 mL) was charged with a solution of keto ester 32 (827 mg, 2.52 mmol) in MeOH (20 mL). After addition of $[(S)$ -BINAP)RuCl₂]₂ · NEt₃ (3.1 mm in THF, 10 mL, 0.031 mmol, 1.3 mol%) the autoclave was pressurized with $H₂$ (80 atm) and the reaction mixture was stirred for 6.5 h at 25° C. After the autoclave had been vented, the solvent was removed in vacuo and the residue was purified by flash chromatography (hexanes/ethyl acetate 4:1) affording product 33 (777 mg, 93%) as a colorless syrup. $[\alpha]_D^{20} = +41.5$ ($c = 1.20$, CH₂Cl₂); ¹H NMR (CD₂Cl₂, 300 MHz): δ = 5.16 – 5.08 (m, 1H), 4.67 (d, J = 6.8 Hz, 1H), 4.60 (d, $J = 6.8$ Hz, 1H), 4.15 – 4.06 (m, 1H), 3.67 – 3.60 (m, $1\,\mathrm{H}$), $3.55 - 3.30$ (br s, $1\,\mathrm{H}$), 3.36 (s, $3\,\mathrm{H}$), 2.41 (dd, $J = 15.7$, $4.6\,\mathrm{Hz}$, $1\,\mathrm{H}$), 2.32 $(dd, J=15.7, 7.9 \text{ Hz}, 1 \text{ H}), 1.98-1.78 \text{ (m, 3 H)}, 1.70 \text{ (s, 3 H)}, 1.69-1.41 \text{ (m,$ 2 H), 1.60 (s, 3 H), 1.44 (s, 9 H), 0.86 (d, $J = 6.4$ Hz, 3 H); ¹³C NMR (CD₂Cl₂, 75.5 MHz : $\delta = 172.0, 132.8, 123.2, 96.0, 81.0, 80.5, 67.8, 56.1, 43.2, 36.8, 36.5,$ $31.7, 28.2, 25.9, 18.0, 14.1; IR (KBr): v = 3468, 2971, 2932, 1729, 1632, 1455,$ 1392, 1368, 1340, 1302, 1258, 1214, 1151, 1097, 1034, 951, 917, 844, 774 cm⁻¹; MS (EI): m/z (%): 330 (-1) [M] , 242 (29), 224 (14), 212 (42), 183 (20), 145 (38), 123 (13), 115 (61), 95 (21), 81 (14), 69 (44), 57 (63), 45 (100); HR-MS (CI): $(C_{18}H_{34}O_5 + H)$ calcd 331.2484; found 331.2485.

 $(3R, 5R, 6S)$ -5-Methoxymethyl-6,9-dimethyl-dec-8-ene-1,3-diol (34) : LiAlH₄ (1.72 g, 45.3 mmol) was added to a solution of compound 33 (5.01 g, 15.1 mmol) in Et₂O (600 mL) at 0 °C. Careful addition of aq. sat. NH₄Cl (100 mL) after 6 h, extraction with Et₂O (3×150 mL), drying of the combined organic phases over Na₂SO₄, evaporation of the solvent and flash chromatography of the residue (hexanes/ethyl acetate $1:1 \rightarrow 1:2$) afforded diol 34 as a colorless syrup (3.56 g, 90%). $\lbrack a \rbrack_{D}^{20} = +51.1$ ($c = 1.19$, CH₂Cl₂); ¹H NMR (CD₂Cl₂, 300 MHz): δ = 5.15 – 5.07 (m, 1H), 4.67 (d, J = 6.6 Hz, 1 H), 4.62 (d, $J = 6.6$ Hz, 1 H), 4.05 – 3.95 (m, 1 H), 3.81 – 3.65 (m, 4 H), 3.37 $(s, 3H), 3.20 - 2.85$ (br s, 1H), $1.98 - 1.78$ (m, 3H), $1.74 - 1.61$ (m, 3H), 1.70 $(s, 3H), 1.59$ $(s, 3H), 1.55 - 1.46$ $(m, 1H), 0.86$ $(d, J = 6.0 \text{ Hz}, 3H);$ ¹³C NMR $(CD_2Cl_2, 75.5 MHz)$: $\delta = 132.9, 123.1, 95.9, 81.9, 72.2, 61.6, 56.2, 39.3, 36.8,$ $36.7, 31.9, 25.8, 17.9, 13.8$; IR (neat): $\nu = 3386, 2961, 2931, 2888, 1658, 1442,$ 1377, 1212, 1151, 1097, 1037, 969, 918, 864, 822, 724 cm⁻¹; MS (EI): m/z (%): 260 (-1) [M] , 228 (5), 210 (2), 198 (8), 183 (10), 165 (2), 141 (7), 124 (8), 110 (14), 101 (43), 95 (17), 83 (17), 69 (34), 55 (29), 45 (100), 41 (25), 29 (12); HR-MS (CI): $(C_{14}H_{28}O_4 + H)$ calcd 261.2066; found 261.2065.

(3S,5S,6R)-1-(4-Methoxybenzyl)-5-methoxymethyl-6,9-dimethyl-dec-8-en-3-ol (35): A solution of diol 34 (2.074 g, 7.96 mmol) in DMF (70 mL) was added to a suspension of NaH (768 mg, 32.0 mmol) in DMF (100 mL). The mixture was stirred for 75 min before PMBCl (1080 µL, 7.96 mmol) was added through a syringe. After stirring for another 90 min, the reaction was quenched with diethylamine (1 mL), the mixture was diluted with ethyl acetate (200 mL) and the organic phase was washed with brine $(3 \times 50 \text{ mL})$. Drying over $Na₂SO₄$, evaporation of the solvent and flash chromatography (hexanes/ethyl acetate 2:1) gave product 35 as a colorless syrup (2.29 g) 76%). $[\alpha]_D^{20} = +26.0$ (c = 1.32, CH₂Cl₂); ¹H NMR (CD₂Cl₂, 300 MHz): δ = 7.25 (d, $J = 8.6$ Hz, 2H), 6.86 (d, $J = 8.6$ Hz, 2H), 5.16 – 5.07 (m, 1H), 4.67 $(d, J=6.7 \text{ Hz}, 1 \text{ H}), 4.60 \text{ (d, } J=6.6 \text{ Hz}, 1 \text{ H}), 4.42 \text{ (s, } 2 \text{ H}), 3.94-3.84 \text{ (m, }$ 1H), 3.79 (s, 3H), 3.69 - 3.53 (m, 3H), 3.50 - 3.25 (br s, 1H), 3.36 (s, 3H), $1.97 - 1.49$ (m, 7H), 1.71 (s, 3H), 1.59 (s, 3H), 0.86 (d, $J = 6.3$ Hz, 3H); ¹³C NMR (CD₂Cl₂, 75.5 MHz): δ = 159.6, 132.7, 131.0, 129.6, 123.3, 114.0, 95.9, 81.1, 73.1, 69.7, 68.3, 56.1, 55.5, 37.5, 37.1, 36.8, 31.8, 25.9, 17.9, 14.1; IR $(KBr): \nu = 3463, 2930, 1613, 1586, 1514, 1463, 1442, 1375, 1302, 1248, 1210,$ 1173, 1152, 1096, 1036, 969, 917, 821, 773, 756, 707, 637, 571, 518 cm⁻¹; MS (EI): m/z (%): 380 (<1) [M]⁺, 348 (1), 247 (1), 227 (2), 197 (4), 176 (2), 151 (1), 137 (13), 121 (100), 101 (7), 69 (7), 55 (4), 45 (15); HR-MS (CI): $(C_{22}H_{36}O_5 + H)$ calcd 381.2641; found 381.2640.

2-Allyl-6-methoxy-benzoic acid (1S,3R,4S)-1-[2-(4-methoxybenzyl)-ethyl]-3-methoxymethyl-4,7-dimethyl-oct-6-enyl ester (36 c): A solution of alcohol 35 (1.582 g, 4.16 mmol) and PPh₃ (1.63 g, 6.21 mmol) in Et₂O (40 mL) was added dropwise to a solution of the carboxylic acid 12 (1.20 g, 6.21 mmol)^[21] and DEAD (970 μ L, 6.21 mmol) in Et₂O (50 mL). After stirring for 20 h, the mixture was concentrated to a volume of ca. 10 mL and precipitated triphenylphosphine oxide was filtered off. Drying of the filtrate over Na₂SO₄, evaporation of the solvent, and flash chromatography (hexanes/ethyl acetate 20:1) afforded ester 36 c as a colorless syrup (2.10 g, 93%). $[\alpha]_D^{20} = +15.1$ (c = 1.02, CH₂Cl₂); ¹H NMR (CD₂Cl₂, 300 MHz): δ = 11.09 (s, 1H), 7.31 (dd, $J = 8.0$, 7.5 Hz, 1H), 7.21 (d, $J = 8.6$ Hz, 2H), 6.84 (d, $J = 8.0$ Hz, 1 H), 6.81 (d, $J = 8.6$ Hz, 2 H), 6.73 (d, $J = 7.5$ Hz, 1 H), 5.98 (ddt, $J = 17.1, 10.2, 6.1 \text{ Hz}, 1 \text{ H}$), $5.62 - 5.53 \text{ (m, 1H)}, 5.12 - 5.05 \text{ (m, 1H)}, 5.01 \text{ (dd,$ $J = 10.2, 1.6$ Hz, 1H), 4.93 (dd, $J = 17.1, 1.6$ Hz, 1H), 4.64 (d, $J = 6.8$ Hz, 1H), 4.55 (d, $J = 6.8$ Hz, 1H), 4.38 (s, 2H), 3.76 (s, 3H), 3.69 (d, $J = 6.0$ Hz, 1H), 3.64 (d, $J = 6.0$ Hz, 1H), 3.60 – 3.49 (m, 3H), 3.33 (s, 3H), 2.12 – 1.48 $(m, 7H), 1.63$ (s, 3H), 1.56 (s, 3H), 0.87 (d, $J=6.4$ Hz, 3H); ¹³C NMR $(CD_2Cl_2, 75.5 MHz)$: $\delta = 171.1, 162.7, 159.6, 143.0, 138.3, 134.3, 132.8, 130.8,$ 129.6, 123.1, 122.7, 116.3, 115.5, 113.9, 113.3, 96.9, 78.5, 73.0, 72.4, 66.8, 55.5, $54.5, 40.3, 37.4, 35.5, 35.2, 31.8, 25.7, 17.9, 13.8; \text{ IR (neat): } \nu = 3374, 3059,$ 2960, 2928, 1723, 1656, 1608, 1578, 1514, 1450, 1374, 1302, 1249, 1222, 1165, 1098, 1039, 999, 917, 818, 767, 712, 573 cm⁻¹; MS (EI): m/z (%): 540 (<1) [M] , 508 (2), 387 (1), 330 (1), 298 (2), 211 (2), 179 (3), 161 (7), 139 (1), 121 (100), 109 (1), 69 (4), 45 (10); HR-MS (ESI pos): $(C_{33}H_{46}O_7 + Na)$ calcd 577.3141; found 577.3143.

(7S,9R,10S)-4-Methoxy-7-[2-(4-methoxybenzyl)-ethyl]-9-methoxymethyl-10-methyl-7,8,9,10,11,14-hexahydro-6-oxa-benzocyclododecen-5-one

(37c): Complex 24 (68 mg, 10 mol%) was added to a solution of diene $36c$ $(R = Me, 450$ mg, 0.811 mmol) in toluene (600 mL) at 80 °C. The reaction was stopped after 90 min upon addition of ethoxy-ethene (5 mL). Evaporation of the solvent and chromatography (hexanes/Et₂O 4:1) on a pre-packed column (LiChroprep Si 60, size A, E. Merck, Darmstadt, Germany) afforded compound (E) -37 c (246 mg, 61%) and (Z)-37 c $(130 \text{ mg}, 32\%)$. (E) -37c: $[\alpha]_D^{20} = -21.0$ $(c=1.0, \text{ benzene})$; ¹H NMR $(C_6D_6, 300 MHz)$: $\delta = 7.40$ (d, $J = 8.6$ Hz, 2H), 7.05 (t, $J = 8.0$ Hz, 1H), 6.91 (d, $J = 8.5$ Hz, 2H), 6.66 (d, $J = 7.6$ Hz, 1H), 6.45 (d, $J = 8.5$ Hz, 1H), $6.01 - 5.90$ (m, 1H), $5.58 - 5.31$ (m, 2H), 5.13 (d, $J = 6.7$ Hz, 1H), 4.91 (d, $J = 6.7$ Hz, 1 H), 4.54 (m, 1 H), 4.42 (s, 2 H), 3.88 – 3.76 (m, 3 H), 3.45 (s, 3 H), 3.41 (s, $3H$), $3.33 - 3.20$ (m, $1H$), 3.28 (s, $3H$), $2.25 - 2.08$ (m, $4H$), $1.84 - 1.67$ (m, 3H), 0.97 (d, $J = 6.0$ Hz, 3H); ¹³C NMR (C₆D₆, 75.5 MHz): $\delta = 168.5$, 160.2, 157.6, 140.1, 132.1, 132.0, 130.3, 129.5, 126.4, 123.5, 97.7, 80.1, 73.3, 72.6, 67.3, 55.9, 55.5, 55.3, 38.6, 38.5, 37.7, 36.9, 34.9, 27.6, 14.1; IR (neat): $\nu =$ 3067, 2956, 2932, 2839, 1724, 1584, 1513, 1468, 1301, 1274, 1249, 1204, 1154, 1117, 1085, 1040, 972 cm⁻¹; MS (EI): m/z (%): 498 (<1) [M]⁺, 300 (10), 189 (6), 187 (7), 176 (10), 121 (100), 95 (3), 45 (26); HR-MS (ESI pos): $(C_{29}H_{38}O_7 + Na)$ calcd 521.2515; found 521.2514. (Z)-37 c: $[\alpha]_D^{20} = -18.5$ $(c = 1.0, \text{ benzene})$; ¹H NMR $(C_6D_6, 300 \text{ MHz})$: $\delta = 7.41 \text{ (d, } J = 8.5 \text{ Hz, } 2 \text{ H}),$ 7.11 (t, $J = 7.7$ Hz, 1 H), 6.92 (d, $J = 8.5$ Hz, 2 H), 6.72 (d, $J = 7.7$ Hz, 1 H), 6.50 (d, $J = 8.5$ Hz, 1H), 5.94 – 5.89 (m, 1H), 5.52 – 5.36 (m, 2H), 4.98 (d, $J = 6.7$ Hz, 1 H), 4.84 (d, $J = 6.7$ Hz, 1 H), 4.45 (s, 2 H), 4.20 – 4.12 (m, 2 H), $3.72 - 3.60$ (m, 2H), 3.42 (s, 3H), 3.39 (s, 3H), 3.36 (s, 3H), $2.23 - 1.98$ (m, 4H), 1.90 – 1.67 (m, 3H), 1.02 (d, $J = 6.6$ Hz, 3H); ¹³C NMR (C₆D₆, 75.5 MHz : $\delta = 167.0, 160.1, 158.1, 140.8, 131.9, 130.9, 130.1, 130.0, 129.5,$

123.4, 114.5, 110.5, 97.7, 78.4, 73.5, 72.6, 67.2, 60.6, 56.1, 55.8, 55.2, 37.2, 37.1, $36.6, 33.4, 32.7, 14.1; IR (neat): $\nu = 3068, 2957, 2933, 1730, 1612, 1598, 1584,$$ 1514, 1469, 1440, 1374, 1265, 1250, 1154, 1115, 1066, 1039 cm⁻¹; MS (EI): m/z (%): 498 (<1) [M]⁺, 330 (5), 317 (8), 300 (10), 189 (7), 176 (12), 121 (100), 45 (27); HR-MS (CI): $(C_{29}H_{38}O_7)$ calcd 499.269; found 499.2697.

(Z)-Hept-2-en-4-ynoic methyl ester (40): A suspension of 1-butynyllithium (0.78 g, 13.0 mmol) in THF (15 mL) was treated at -50° C with a solution of anhydrous zinc chloride (2.04 g, 15.0 mmol) in THF (20 mL). The resulting mixture was allowed to warm to ambient temperature. Ester 39 (2.12 g, 10.0 mmol) and a solution of $[Pd(PPh₃)₄]$ (232 mg, 0.2 mmol, 2 mol%) in THF (20 mL) were subsequently added. The reaction mixture was stirred for 3 h and quenched by the addition of water. The aqueous layer was repeatedly extracted with $Et₂O$, the combined organic phases were washed with brine, dried over $Na₂SO₄$ and evaporated. Flash chromatography (pentane/Et₂O 30:1) afforded product 40 (1.24 g, 90%) as a colorless liquid. ¹H NMR (CD₂Cl₂, 300 MHz): δ = 6.13 (dt, J = 11.5, $2.3 \text{ Hz}, 1 \text{ H}$), 6.01 (d, J = 11.5 Hz, 1H), 3.71 (s, 3H), 2.43 (qd, J = 7.5, 2.3 Hz, 2H), 1.19 (t, J = 7.5 Hz, 3H); ¹³C NMR (CD₂Cl₂, 75.5 MHz): δ = 165.4, $127.4, 124.2, 105.4, 77.3, 51.5, 14.0, 13.7; \text{ IR} \text{ (neat): } \nu = 3018, 2980, 2950,$ 2918, 2880, 2211, 1730, 1717, 1611, 1437, 1404, 1291, 1234, 1195, 1177, 817 cm⁻¹; MS (EI): m/z (%): 138 (100) [*M*]⁺, 137 (28), 123 (31), 110 (14), 107 (48), 95 (36), 79 (38), 77 (41), 67 (37), 78 (12), 77 (41), 67 (37), 63 (13), 51 (20), 39 (12); HR-MS (EI): $(C_8H_{10}O_2)$ calcd 138.0681; found 138.0681; elemental analysis calcd for $C_8H_{10}O_2$ (138.17): C 69.55, H 7.30; found C 69.50, H 7.26.

(2Z,4Z)-Hepta-2,4-dienoic methyl ester (41): Quinoline (21 μ L, 8 mol%) and alkyne 40 (291 mg, 2.109 mmol) were dissolved in $CH₂Cl₂$ (30 mL). Commercially available Lindlar catalyst (105 mg, 3 mol%) was added and the resulting suspension was stirred for 50 min under an atmosphere of H_2 (1 atm). The catalyst was filtered off through a pad of Celite, the solvent was evaporated and the residue was purified by flash chromatography (pentane/Et₂O 30:1) affording diene 41 (237 mg, 80%) as a colorless liquid. ¹H NMR (CD₂Cl₂, 300 MHz): δ = 7.21 (dd, J = 11.4, 1.2 Hz, 1H), 6.95 (dd, $J = 11.7, 0.8$ Hz, 1 H), 5.98 – 5.86 (m, 1 H), 5.66 (d, $J = 11.4$ Hz, 1 H), 3.69 (s, 3H), 2.36–2.22 (m, 2H), 1.03 (t, $J=7.5$ Hz, 3H); ¹³C NMR (CD₂Cl₂, 75.5 MHz : $\delta = 167.1, 143.6, 139.2, 124.0, 117.2, 51.3, 21.2, 14.1; \text{ IR (neat)}$: - 3053, 2968, 2876, 1720, 1631, 1592, 1444, 1365, 1231, 1194, 1175, 1132 cm⁻¹; MS (GC-EI): *m*/z (%): 140 (17) [*M*]⁺, 111 (100), 109 (19), 81 (48), 80 (12), 79 (39), 53 (20), 41 (18), 39 (20), 27 (10); HR-MS (EI): $(C_8H_{12}O_2)$ calcd 140.0837; found 140.0837.

(2Z,4Z)-Hepta-2,4-dienoic acid amide (42): Compound 41 (300 mg, 2.141 mmol) was added to aq. ammonium hydroxide (50 mL, 30%). The reaction mixture was stirred at ambient temperature for 4 days. The resulting mixture was extracted with ethyl acetate, the combined organic layers were dried over Na_2SO_4 and evaporated. The residue was purified by flash chromatography (ethyl acetate) to afford amide 42 (167 mg, 62%) as a white solid. ¹H NMR (CDCl₃, 300 MHz): δ = 7.22 – 7.12 (m, 1 H), 6.77 (dd, $J = 11.6, 1.1$ Hz, 1 H), 5.95 – 5.55 (br s, 2 H), 5.88 – 5.77 (m, 1 H), 5.62 (d, $J =$ 11.5 Hz, 1 H), 2.23 (dtd, $J = 15.2, 7.5, 1.6$ Hz, 2 H), 1.02 (t, $J = 7.5$ Hz, 3 H); ¹³C NMR (CDCl₃, 75.5 MHz): δ = 168.6, 141.9, 136.2, 123.6, 119.3, 20.7, 13.9; IR (KBr): ν = 3403, 3198, 3009, 2967, 2933, 1653, 1607, 1591, 1457, 1327, 808 cm⁻¹; MS (GC-EI): *m*/z (%): 125 (6) [*M*]+, 110 (3), 96 (100), 81 (15), 67 (12), 53 (12), 41 (17), 27 (12); elemental analysis calcd for $C_7H_{11}NO$ (125.17): C 67.17, H 8.86, N 11.19; found C 67.26, H 8.81, N 11.12.

(7S,9R,10S,12E)-7-(2-Hydroxyethyl)-4-methoxy-9-methoxymethoxy-10 methyl-7,8,9,10,11,14-hexahydro-6-oxa-benzocyclododecen-5-one (47): A solution of compound $37c$ (420 mg, 0.843 mmol) and DDQ (239 mg, 1.053 mmol) in water (1 mL) and CH_2Cl_2 (18 mL) was stirred at ambient temperature for 12 h. The reaction was diluted with ethyl acetate (40 mL) and filtered through a pad of Celite, the filtrate was evaporated and the residue was purified by flash chromatography (hexane/ethyl acetate 1:1) affording compound 47 as a colorless syrup (300 mg, 94%). $[\alpha]_D^{20} = -34.5$ $(c=1.0, CHCl₃)$; ¹H NMR $(CD₂Cl₂, 300 MHz)$: $\delta = 7.14$ (t, $J = 7.7$ Hz, 1H), 6.73 (d, J = 7.7 Hz, 1 H), 6.68 (d, J = 7.7 Hz, 1 H), 5.36 – 5.30 (m, 2 H), 5.25 – 5.20 (m, 2H), 4.68 (q, $J = 6.7$ Hz, 2H), 4.02 – 3.93 (m, 2H), 3.69 (s, 2H),

 $3.68 - 3.52$ (m, 2H), 3.30 (s, 3H), $3.25 - 3.22$ (m, 1H), $1.79 - 1.58$ (m, 4H), $1.36 - 1.27$ (m, 2H), $1.15 - 1.06$ (m, 2H), 0.74 (d, $J = 6.6$ Hz, 3H); ¹³C NMR $(CD_2Cl_2, 75.5 MHz)$: $\delta = 168.5, 156.6, 139.5, 131.8, 130.4, 128.8, 125.0, 123.4,$ 109.7, 97.0, 79.6, 73.1, 59.5, 56.0, 55.7, 39.5, 38.1, 38.0, 36.2, 34.4, 13.5; IR $(neat):$ $\nu = 3438, 3068, 2956, 2930, 2843, 1723, 1597, 1584, 1468, 1439, 1275,$ 1119, 1085, 1071, 1038, 972 cm⁻¹; MS (EI): m/z (%): 378 (3) [M]⁺, 346 (9),

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FULL PAPER A. Fürstner et al.

316 (12), 315 (10), 260 (9), 259 (13), 228 (16), 215 (21), 204 (12), 189 (18), 187 (49), 186 (13), 177 (11), 175 (24), 174 (20), 163 (15), 162 (39), 161 (27), 159 (13), 153 (11), 148 (12), 115 (16), 55 (18), 45 (100), 43 (13); HR-MS (ESI pos): $(C_{21}H_{30}O_6 + Na)$ calcd 401.1940; found 401.1942.

(7S,9R,10S,12E)-(4-Methoxy-9-methoxymethoxy-10-methyl-5-oxo-7,8,9,10,11,14-hexahydro[5H]-6-oxa-benzocyclododecen-7-yl)-acetalde-

hyde (48): Dess-Martin periodinane (403 mg, 0.950 mmol) was added at 0° C to a solution of alcohol 47 (300 mg, 0.793 mmol) in CH₂Cl₂ (60 mL). The solution was allowed to warm to room temperature and stirred for 18 h. The reaction was diluted with ethyl acetate (50 mL) and quenched by the addition of sat. aq. NaHCO₃ (50 mL) and sat. aq. Na₂S₂O₃ (50 mL). The mixture was vigorously stirred until a clear solution resulted. The aqueous layer was repeatedly extracted with ethyl acetate, the combined organic phases were washed with brine, dried over $Na₂SO₄$ and evaporated. Flash chromatography (hexane/ethyl acetate 4:1) furnished aldehyde 48 (260 mg, 87%) as a colorless syrup. $[\alpha]_D^{20} = -42.7$ ($c = 2.0$, benzene); ¹H NMR $(C_6D_6, 300 MHz)$: $\delta = 9.65$ (dd, $J = 4.0, 0.8 \text{ Hz}, 1 \text{ H}$), 7.00 (t, $J = 7.8 \text{ Hz}, 1 \text{ H}$), 6.62 (d, $J = 7.8$ Hz, 1 H), 6.38 (d, $J = 7.8$ Hz, 1 H), 6.11 (m, 1 H), 5.41 (m, 1 H), 5.10 (d, $J = 6.8$ Hz, 1H), 4.94 (d, $J = 6.8$ Hz, 1H), 4.50 (m, 1H), 4.01 (q, 7.3 Hz, 1H), 3.84 (m, 1H), 3.42 (s, 3H), 3.39 (s, 3H), 2.52 (m, 1H), 2.23 ± 2.06 (m, 3H), 1.75 – 1.65 (m, 2H), 1.43 – 1.33 (m, 2H), 0.93 (d, $J = 6.6$ Hz, 3H); ¹³C NMR (C₆D₆, 75.5 MHz): δ = 199.9, 168.4, 156.7, 139.8, 132.1, 130.5, 129.6, 123.2, 109.9, 97.9, 80.0, 70.2, 55.9, 55.5, 49.9, 38.5, 38.4, 36.5, $35.1, 27.6, 14.0; \text{IR} \text{ (neat): } \nu = 3497, 3068, 2957, 2930, 2733, 1728, 1597, 1584,$ 1469, 1439, 1274, 1254, 1147, 1116, 1085, 1037, 971 cm⁻¹; MS (EI): m/z (%): 376 (3) [M] , 314 (22), 259 (17), 228 (11), 215 (18), 188 (10), 187 (47), 186 (13), 175 (11), 174 (14), 163 (11), 162 (13), 161 (14), 159 (12), 148 (11), 115 (13), 55 (11), 45 (100); HR-MS (ESI pos): $(C_{21}H_{28}O_6 + Na)$ calcd 399.1784; found 399.1783.

(7S,9R,10S,12E)-7-(3-Iodoallyl)-4-methoxy-9-methoxymethoxy-10-methyl-7,8,9,10,11,14-hexahydro-6-oxa-benzocyclododecen-5-one (49): To a slurry of flame-dried CrCl₂ (1018 mg, 8.283 mmol) in THF (5 mL) was added a solution of aldehyde 48 (260 mg, 0.691 mmol) and iodoform (1.10 g, 2.794 mmol) in 1,4-dioxane (30 mL). The resulting brown solution was stirred at ambient temperature for 16 h, diluted with $Et₂O$ (40 mL), and poured into water (ca. 40 mL). The aqueous layer was separated, saturated with NaCl, and extracted with $Et₂O$ (50 mL, in several portions). The combined organic phases were washed with brine, dried over $Na₂SO₄$, filtered, and evaporated. Purification of the residue by flash chromatography (hexanes/ethyl acetate 10:1) gave product 49 (302 mg, 87%) as a syrup (9:1 mixture of *E*:*Z* isomers). $[a]_D^{20} = -37.5$ (*c*=1.0, benzene); ¹H NMR (C₆D₆, 300 MHz): δ = 7.05 (t, J = 7.6 Hz, 1H), 7.01 – 6.89 (m, 1H), 6.65 (d, $J = 7.6$ Hz, 1H), 6.47 (d, $J = 7.6$ Hz, 1H), 5.89 (d, $J = 5.5$ Hz, 1H), 5.57 (m, 1H), 5.49 – 5.39 (m, 1H), 5.08 (d, $J = 6.8$ Hz, 1H), 4.88 (d, $J =$ 6.8 Hz, 1 H), 4.46 (d, $J = 6.8$ Hz, 1 H), 3.90 (q, $J = 7.0$ Hz, 1 H), 3.85 (m, 1 H), 3.53 (s, 3H), 3.40 (s, 3H), 2.36 - 2.07 (m, 3H), 1.79 - 1.32 (m, 5H), 0.95 (d, $J = 6.6$ Hz, 3H); ¹³C NMR (C₆D₆, 75.5 MHz): $\delta = 168.4$, 157.7, 143.2, 139.9, 132.0, 130.4, 129.4, 123.4, 110.2, 97.8, 97.7, 80.0, 77.9, 73.8, 56.3, 55.9, 43.2, $38.5, 38.4, 34.9, 35.1, 14.0; \text{IR} \text{ (neat): } v = 3425, 2956, 2929, 2840, 1726, 1658,$ 1597, 1584, 1468, 1438, 1424, 1274, 1117, 1085, 1040, 972; MS (EI): m/z (%): 500 (5) [M] , 455 (22), 438 (19), 307 (12), 275 (10), 260 (13), 259 (34), 245 (15), 229 (10), 228(13), 227 (18), 215 (24), 213 (11), 204 (10), 199 (13), 189 (11), 187 (41), 186 (13), 185 (11), 177 (11), 175 (13), 174 (12), 173 (10), 163 (17), 162 (13), 161 (23), 159 (13), 148 (10), 115 (15), 91 (11), 67 (12), 55 (13), 45 (100); elemental analysis calcd for $C_{22}H_{29}IO_5$ (500.38): C 52.81, H 5.84; found C 52.70, H 5.77.

(7S,9R,10S,12E)-4,9-Dihydroxy-7-(3-iodoallyl)-10-methyl-7,8,9,10,11,14-

hexahydro-6-oxa-benzocyclododecen-5-one (50): A solution of BBr_3 (1M in CH_2Cl_2 , 450 µL) was added dropwise at -78° C to a solution of compound 49 (90 mg, 0.179 mmol) in CH_2Cl_2 (20 mL). The resulting brownish red solution was stirred at that temperature for 1 h and allowed to warm to room temperature. The reaction was quenched by the addition of water (50 mL). The aqueous layer was extracted with CH_2Cl_2 , the organic extracts were washed with brine, dried over $Na₂SO₄$ and evaporated. Purification of the residue by flash chromatography (hexanes/ethyl acetate 4:1) afforded product **50** (70 mg, 88%) as a colorless wax. $[\alpha]_D^{20} = -31.5$ $(c=1.0, \text{ benzene})$; ¹H NMR (CD₃OD, 300 MHz): $\delta = 7.20$ (t, $J = 7.6$ Hz, $1\,\mathrm{H}$), 6.91 – 6.37 (m, 3 $\,\mathrm{H}$), 6.36 (d, J = 14.5 $\,\mathrm{Hz}$, 1 $\,\mathrm{H}$), 5.44 – 5.40 (m, 3 $\,\mathrm{H}$), 4.92 $(s, 2H)$, 4.25 (d, $J = 8.5$ Hz, 1H), 3.72 – 3.62 (m, 2H), 2.58 – 2.48 (m, 3H), $1.89-1.81$ (m, 3H), 1.42 (m, 1H), 0.94 (d, $J=6.6$ Hz, 3H); ¹³C NMR $(CD_3OD, 75.5 MHz)$: $\delta = 172.7, 158.5, 145.4, 142.3, 140.6, 133.7, 133.2, 132.2,$

125.1, 124.1, 116.9, 79.4, 76.8, 73.7, 44.9, 40.7, 40.2, 38.9, 15.2; IR (neat): $\nu =$ 3433, 2958, 2914, 1691, 1652, 1605, 1589, 1465, 1295, 1268, 1249, 1216, 1126, 1066, 1028, 968; MS (EI): m/z (%): 442 (12) [M] , 425 (12), 424 (17), 295 (10), 276 (14), 275 (12), 257 (14), 251 (25), 239 (11), 232 (31), 231 (519), 230 (14), 213 (17), 203 (11), 202 (14), 201 (18), 192 (14), 191 (24), 190 (20), 185 (15), 176 (17), 175 (32), 175 (38), 173 (68), 172 (100), 167 (13), 163 (11), 162 (16), 161 (27), 160 (23), 159 (15), 152 (11), 151 (14), 150 (14), 149 (29), 148 (33), 147 (25), 145 (22), 144 (11), 135 (11), 134 (17), 133 (10), 132 (11), 131 (15), 128 (13), 127 (22), 117 (11), 116 (20), 115 (28), 109 (11), 107 (14), 105 (17), 95 (19), 91 (17), 83 (23), 81 (21), 79 (17), 78 (10), 77 (17), 69 (16), 68 (13), 67 (31), 66 (21), 65 (11), 57 (17), 55 (40), 53 (14), 43 (31), 41 (41), 39 (19), 29 (19); HR-MS (EI): $(C_{19}H_{23}IO_4)$ calcd 442.0641; found 442.0637. Salicylihalamide (1a, b): An oven-dried Schlenk flask was charged with amide 42 (25 mg, 0.202 mmol), CuTC (13 mg, 0.068 mmol) and Rb_2CO_3 (46 mg, 0.202 mmol). Anhydrous DMA (1 mL) was added, the resulting suspension was carefully degassed in vacuo, compound 50 (30 mg, 0.068 mmol) was introduced, and degassing was repeated prior to heating the mixture to 90° C for 2 h under argon. The red slurry was cooled to room temperature and diluted wit Et_2O , the organic phase was washed with pH 7 buffer, the aqueous layer was extracted twice with $Et₂O$, the combined organic phases were dried over Na₂SO₄ and evaporated. The residue was purified by preparative HPLC (Shimadzu LC-8; column: 125/20 mm $BIAX$; eluent: MeOH/H₂O 7:3) giving both isomers of the title compound (17 mg, 57%, $1a:1b \approx 2.5:1$) as colorless solids. Spectroscopic data of $1a$: $[\alpha]_{\text{D}}^{20} = -33.0 \; (c = 1.0, \text{ benzene})$; ¹H NMR (CD₃OD, 300 MHz): $\delta = 7.34$ (t, $J = 10.8$ Hz, 1H), 7.17 (t, $J = 7.5$ Hz, 1H), 6.91 (t, $J = 11.8$ Hz, 1H), 6.87 (d, $J = 13.3$ Hz, 1 H), 6.77 (d, $J = 7.8$ Hz, 1 H), 6.70 ($J = 7.5$ Hz, 1 H), 5.85 (m, 1 H), 5.73 (d, $J = 11.5$ Hz, 1 H), 5.52 – 5.34 (m, 4 H), 4.15 (dd, $J = 8.7$, 3.3 Hz, $1\,\mathrm{H}$), 3.59 (dd, $J = 16.4, 7.3 \,\mathrm{Hz}$, $1\,\mathrm{H}$), 3.39 (m, $1\,\mathrm{H}$), 2.46 (m, $1\,\mathrm{H}$), 2.37 – 2.28 $(m, 4H), 1.97 - 1.74$ $(m, 3H), 1.41$ $(m, 1H), 1.07$ $(t, J = 7.5$ Hz, $3H), 0.90$ $(d,$ $J = 6.7$ Hz, 3H); ¹³C NMR (CD₃OD, 75.5 MHz): $\delta = 172.3, 167.1, 158.4,$ 143.8, 141.9, 141.2, 138.9, 132.9, 132.8, 131.9, 127.4, 126.5, 123.6, 121.5, 116.5, 111.6, 77.2, 73.2, 40.1, 40.0, 39.7, 38.7, 37.8, 22.7, 15.5, 14.8; MS (EI) m/z (%) 439 (14) [M] , 192 (20), 191 (19), 173 (9), 149 (9), 127 (9), 109 (100), 108 (13), 83 (27), 82 (33), 81 (73), 79 (20), 57 (12), 56 (15), 55 (15), 53 (13), 41 (17); spectroscopic data of **1b**: $[\alpha]_D^{20} = -65.0$ ($c = 0.5$, benzene); ¹H NMR $(C_6D_6, 300 MHz)$: $\delta = 8.03$ (t, $J = 11.8$ Hz, 1H), 7.86 (d, $J = 11.2, 1$ H), 7.38 $(t, J = 10.5 \text{ Hz}, 1 \text{ H}), 7.04 \text{ (m, 1 H)}, 6.73 \text{ (t, } J = 11.6 \text{ Hz}, 1 \text{ H}), 6.57 \text{ (dd}, J = 7.0,$ $3.0 \text{ Hz}, 1 \text{ H}$), $5.73 \text{ (m}, 1 \text{ H})$, $5.59 \text{ (d}, J = 11.4, 1 \text{ H})$, $5.45 - 5.17 \text{ (m}, 3 \text{ H})$, 4.69 $(q, J = 8.2 \text{ Hz}, 1 \text{ H}), 3.64 \text{ (dd, } J = 16.6, 5.1 \text{ Hz}, 1 \text{ H}), 3.53 \text{ (d, } J = 7.7 \text{ Hz}, 1 \text{ H}),$ 3.43 (d, $J = 16.5$ Hz, 1 H), $2.30 - 2.00$ (m, 4H), $1.91 - 1.73$ (m, 3H), 1.65 (m, 1 H), 1.45 (s, 1 H), 1.34 (dd, $J = 15.0$, 8.0 Hz, 1 H), 0.93 (d, $J = 6.8$ Hz, 3 H), 0.87 (m, 3H). The analytical data are in full agreement with those previously reported.[1, 9]

Enamide formation–Model reactions

(2Z,4Z)-Heptadienoic acid [2-(3-methoxyphenyl)-vinyl]-amide (44 a): An oven-dried Schlenk flask was charged with amide 42 (43 mg, 0.345 mmol), CuTC (7 mg, 0.035 mmol) and $Rb₂CO₃$ (32 mg, 0.138 mmol). Anhydrous DMA (1 mL) was added, the resulting suspension was degassed, vinyl iodide 43a (18 mg, 0.069 mmol) was introduced and the degassing procedure was repeated prior to heating the mixture to 90° C for 3 h under argon. The red slurry was cooled to room temperature, diluted with $Et₂O$ and washed with pH 7 buffer. The aqueous phase was extracted twice with $Et₂O$, the combined organic layers were dried over $Na₂SO₄$ and evaporated. The residue was purified by flash chromatography (hexanes/ethyl acetate 10:1) affording compound 44 a (16 mg, 89%) as a colorless oil. Mixture of isomers $(E.Z=2:1)$; data of the major isomer: ¹H NMR $(CD_2Cl_2,$ 300 MHz : $\delta = 7.63 - 7.51 \text{ (m, 2H)}$, $7.39 - 7.31 \text{ (m, 1H)}$, 7.19 (t, 1H) , $6.95 -$ 6.87 (m, 2H), 6.72 (d, J = 7.9 Hz, 1 H), 6.12 (d, J = 14.1 Hz, 1 H), 5.92 (q, J = $9.6 \text{ Hz}, 1 \text{ H}$), $5.64 \text{ (d, } J = 11.3 \text{ Hz}, 1 \text{ H})$, $5.53 \text{ (d, } J = 11.3 \text{ Hz}, 1 \text{ H})$, 3.79 (s, 3H), 2.32–2.20 (m, 2H), 1.01 (t, $J=7.5$ Hz, 3H); ¹³C NMR (CD₂Cl₂, 75.5 MHz : $\delta = 163.7, 160.6, 143.4, 138.4, 138.0, 130.1, 124.4, 123.6, 119.3,$ 118.7, 112.9, 112.8, 111.1, 55.7, 21.3, 14.3; IR (neat): $\nu = 3273$, 3179, 3070, 2999, 2964, 2932, 2873, 1640, 1596, 1523, 1492, 1464, 1434, 1309, 1261, 1217, 1190, 1158, 1047, 954, 863, 810, 776, 712, 689 cm⁻¹; MS (EI): m/z (%): 257 (60) $[M]^+$, 228 (9) , 149 (100) , 109 (91) , 81 (68) .

(2Z,4Z)-Heptadienoic acid [2-(3-hydroxyphenyl)-vinyl]-amide (44 b): The reaction was performed as described above using amide 42 (38 mg, 0.305 mmol), CuTC (5 mg, 0.030 mmol), Rb_2CO_3 (42 mg, 0.183 mmol), and $3-(2-iodo-vinyl)$ -phenol $43b$ (15 mg, 0.061 mmol). A standard extractive work-up followed by flash chromatography (hexanes/ethyl acetate 4:1)

afforded compound $44b$ (13 mg, 86%) as a colorless syrup. ¹H NMR $(CD_2Cl_2, 300 MHz)$: $\delta = 7.54 - 7.46$ (m, 2H), 7.31 (m, 1H), 7.07 (t, J = 7.7 Hz, $1\,\mathrm{H}$), 6.89 – 6.76 (m, 2 H), 6.60 (d, J = 7.7 Hz, 1 H), 6.03 (d, J = 14.2 Hz, 1 H), 5.87 (q, $J = 7.7$ Hz, 1H), 5.76 (s, 1H), 5.61 (d, $J = 11.5$ Hz, 1H), 2.20 (m, 2H), 1.12 (t, $J = 6.5$ Hz, 3H); ¹³C NMR (CD₂Cl₂, 75.5 MHz): $\delta = 163.9$, 156.7, 143.4, 138.3, 138.0, 130.2, 124.2, 123.3, 119.0, 118.6, 114.1, 112.9, 112.2, $21.2, 14.1; \text{IR} \text{ (neat): } \nu = 3281, 3074, 2967, 2933, 2873, 1666, 1640, 1591, 1526,$ 1492, 1262, 1217, 1159, 950, 869, 778, 689 cm⁻¹; MS (EI): m/z (%): 243 (58) [M] , 214 (8), 135 (79), 109 (100), 108 (10), 81 (63), 79 (16), 53 (17), 41 (13).

(10E)-11-Benzoylamino-undecenoic isopropyl ester (46): An oven-dried Schlenk flask was charged with benzoic acid amide (51 mg, 0.421 mmol), CuTC (16 mg, 0.084 mmol) and Cs_2CO_3 (138 mg, 0.421 mmol). Anhydrous NMP (1 mL) was added and the suspension formed was carefully degassed. Compound 45 (100 mg, 0.282 mmol) was added, the mixture was degassed again prior to heating to 90° C for 4 h under argon. The red slurry was cooled to ambient temperature, diluted with $Et₂O$ and washed with $pH 7$ buffer. The aqueous layer was extracted twice with $Et₂O$, the combined organic layers were dried over $Na₂SO₄$ and concentrated in vacuo. The residue was purified by flash chromatography (hexanes/ethyl acetate 10:1) to afford compound 46 (65 mg, 70%) as a colorless syrup. ¹H NMR $(CD_2Cl_2, 300 MHz)$: $\delta = 7.79$ (d, $J = 8.1$ Hz, 3H), 7.55 – 7.42 (m, 3H), 6.93 $(dd, J=14.2, 10.3 Hz, 1H), 5.32 (dd, J=14.2 Hz, 7.1 Hz, 1H), 4.95 (sept,$ $J = 6.2$ Hz, 1 H), 2.23 (t, $J = 7.2$ Hz, 2 H), 2.06 (q, $J = 7.2$ Hz, 2 H), 1.60 – 1.30 $(m, 10H)$, 1.20 $(d, J = 6.2 \text{ Hz}, 6H)$, 0.92 – 0.87 $(m, 2H)$; ¹³C NMR (CD_2Cl_2) , 75.5 MHz): $\delta = 173.5, 164.3, 134.4, 132.1, 129.0, 127.3, 123.1, 114.5, 67.5, 35.0,$ $30.3, 30.2, 29.7, 29.6, 29.4, 29.3, 25.4, 24.0, 22.0; \text{IR} \text{ (neat): } v = 3310, 2197,$ 3066, 2927, 2854, 1731, 1677, 1640, 1579, 1527, 1489, 1374, 1322, 1259, 1190, 1109, 956, 707 cm⁻¹; MS (EI): m/z (%): 345 (4) [M]⁺, 240 (14), 122 (29), 105 (100), 77 (22); HR-MS (ESI pos): $(C_{21}H_{31}NO_3)$ calcd 345.2304; found 345.2301.

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FULL PAPER A. Fürstner et al.

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